Types of Radioactive Decay Assignment and Quiz

Identify whether each reaction represents a chemical reaction or a nuclear reaction by filling in the blank with the word 'chemical' or 'nuclear'.

Cl + H -> 2HCl

This represents a _____ reaction.

H + H -> He + n

This represents a _____ reaction. - ✓ ✓ chemical nuclear

Which statement best describes the two reactions? - ✓ ✓ Reaction B involves a greater change and a change in element identity.

Explain why isotopes of the same element behave differently in nuclear reactions but not in chemical reactions. - $\sqrt{\cdot}$ Isotopes in chemical reactions produce the same chemical compounds and show similar behavior Isotopes in nuclear reactions show different behavior and produce different products.

Use the drop-down menus to complete the corresponding cells in the table to the right.

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Brendan G. Carr

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