

WKS 7.1 – Beginning Naming & Formula Writing (2 pages)

Name the following ionic compounds. Be sure to check for Roman Numerals!

Na_3P		MgBr_2	
Ag_2Se		AlCl_3	
Ca_3P_2		Ba_3N_2	
Li_2O		ZnF_2	
K_2S		SrI_2	
$(\text{NH}_4)_2\text{S}$		$\text{Ca}_3(\text{PO}_3)_2$	
Li_2CO_3		$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$	
KCN		AgNO_3	
$\text{Sb}(\text{NO}_3)_3$		NiS	
$\text{Cu}(\text{NO}_3)_2$		FeCl_2	
$\text{Au}(\text{CN})_3$		$\text{Pb}(\text{SO}_4)_2$	

Write the correct formulas for each of the ionic compounds.

silver iodide		lithium sulfide	
magnesium nitride		cadmium bromide	
zinc chloride		barium phosphide	
aluminum fluoride		strontium oxide	
potassium permanganate		strontium sulfate	
cesium bromate		magnesium cyanide	
aluminum hypiodite		cadmium periodate	
tin (IV) chlorate		bismuth (III) carbonate	
copper (I) sulfide		iron (III) oxide	

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JA Banks



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Table of Contents Pre Ap Chemistry Naming And Formula Writing Answer

1. Understanding the eBook Pre Ap Chemistry Naming And Formula Writing Answer
 - The Rise of Digital Reading Pre Ap Chemistry Naming And Formula Writing Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Pre Ap Chemistry Naming And Formula Writing Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pre Ap Chemistry Naming And Formula Writing Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pre Ap Chemistry Naming And Formula Writing Answer
 - Personalized Recommendations
 - Pre Ap Chemistry Naming And Formula Writing Answer User Reviews and Ratings
 - Pre Ap Chemistry Naming And Formula Writing Answer and Bestseller Lists

5. Accessing Pre Ap Chemistry Naming And Formula Writing Answer Free and Paid eBooks
 - Pre Ap Chemistry Naming And Formula Writing Answer Public Domain eBooks
 - Pre Ap Chemistry Naming And Formula Writing Answer eBook Subscription Services
 - Pre Ap Chemistry Naming And Formula Writing Answer Budget-Friendly Options
6. Navigating Pre Ap Chemistry Naming And Formula Writing Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Pre Ap Chemistry Naming And Formula Writing Answer Compatibility with Devices
 - Pre Ap Chemistry Naming And Formula Writing Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pre Ap Chemistry Naming And Formula Writing Answer
 - Highlighting and Note-Taking Pre Ap Chemistry Naming And Formula Writing Answer
 - Interactive Elements Pre Ap Chemistry Naming And Formula Writing Answer
8. Staying Engaged with Pre Ap Chemistry Naming And Formula Writing Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pre Ap Chemistry Naming And Formula Writing Answer
9. Balancing eBooks and Physical Books Pre Ap Chemistry Naming And Formula Writing Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pre Ap Chemistry Naming And Formula Writing Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pre Ap Chemistry Naming And Formula Writing Answer
 - Setting Reading Goals Pre Ap Chemistry Naming And Formula Writing Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pre Ap Chemistry Naming And Formula Writing Answer
 - Fact-Checking eBook Content of Pre Ap Chemistry Naming And Formula Writing Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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