Colligative Properties Worksheet

FYI: K₁ of water is 1.86° C/m; K₅ of water is 0.52° C/m

 Explain why the addition of a solute decreases the melting point of a liquid.

What is the melting point of a solution in which 3.5 grams of sodium chloride is added to 230 mL of water?

What is the boiling point of a solution in which 22.5 grams of sucrose (C₁₂H₂₂O₁₁) is dissolved in 550 mL of water?

4) Rank the following solutions from lowest to highest boiling point, a 2.5 m solution of sodium chloride, a 3.5 m solution of magnesium chloride, or a 4.5 m solution of sulfur dioxide?

5) If an aqueous solution has a boiling point of 102.5°, what's the effective molality of the solute?

Practice Test On Colligative Properties

M Lipman

Practice Test On Colligative Properties:

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