

Coulomb's Law

1. The diagram is of a hydrogen atom.
- Label the proton in the nucleus with a + sign and the orbital electron with a - sign.
 - The electrical interaction between the nucleus and the orbital electron is a force of _____ (attractive) _____ (repulsive).
 - According to Coulomb's law,

$$F = k \frac{q_1 q_2}{r^2}$$

if the charge of either the nucleus or the orbital electron were greater, the force between the nucleus and the electron would be _____ (greater) _____ (less).

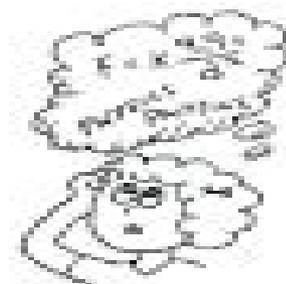
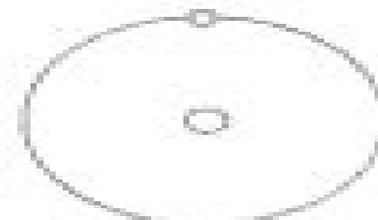
_____ (greater) _____ (less).

and if the distance between the nucleus and electron were greater the force would be _____ (greater) _____ (less).

If the distance between the nucleus and electron were doubled, the force would be _____ (1/4 as much) _____ (1/2 as much) _____ (two times as much) _____ (4 times as much).

2. Consider the electric force between a pair of charged particles a certain distance apart. By Coulomb's law:

- If the charge on one of the particles is doubled, the force is _____ (unchanged) _____ (halved) _____ (doubled) _____ (quadrupled).
- If instead the charge on both particles is doubled, the force is _____ (unchanged) _____ (halved) _____ (doubled) _____ (quadrupled).
- If instead the distance between the particles is halved, the force is _____ (unchanged) _____ (halved) _____ (doubled) _____ (quadrupled).
- If the distance is halved and the charge of both particles is doubled, the force is _____ times as great.



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Practice Page 32 1:

The book delves into Practice Page 32 1. Practice Page 32 1 is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Practice Page 32 1, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Practice Page 32 1
- Chapter 2: Essential Elements of Practice Page 32 1
- Chapter 3: Practice Page 32 1 in Everyday Life
- Chapter 4: Practice Page 32 1 in Specific Contexts
- Chapter 5: Conclusion

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