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# ZI DAW අධ්යයන පොද සහනිත පතු (සංමානය පෙළ) විභාකය, 2021 දෙසැම්බර් General Certificate of Education (Ord. Level) Examination, December 2021

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සියළු ම හිමිකම ඇවරිණි www.mathematics.lk

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ದರ್ಭಾಥವೇ.

- \* සියලු ම පුන්නවලට පිළිතුරු සපයන්න.
- 💌 අංක 1 සිට 40 තෙක් පුන්නවල, පිළිතුර සඳහා (1), (2), (3), (4) ලෙස වරණ හතර බැහින් දී ඇත. එක් එක් පුන්නය සදහා නිවැරදී හෝ වඩාක් ගැළපෙන හෝ පිළිතුරට අදාළ වරණය තෝරා ගන්න.
- \* ඔබට සැපයෙන පිළිතුරු පතුයේ එක් එක් පුන්නය සදහා ඔබ තෝරාගත් වරණයෙහි අංකයට සැසදෙන කවය තුළ (×) උකුණ යොදන්න.
- \* එම පිළිතුරු පතුයේ පිටුපස දී ඇති අනෙක් උපදෙස් ද සැලකිල්ලෙන් කියවා, ඒවා ද පිළිපදින්න.
- බීජ භට ගන්නා අපුෂ්ප ශාකයකට උදාහරණයක් වන්නේ පහත සදහන් කුමන ශාකය ද?
  - (1) මාකාලන්ට්යා
- (2) පෙලසිපොල්ලා

(3) ⊗€

(4) සැල්වීනියා

- 2. සණත්වයේ ඒකකය කමක් ද?
  - (1)  $kg m s^{-2}$

 $(2) \text{ g m}^{-3}$ 

(3) kg m<sup>-2</sup>

4) kg m<sup>-3</sup>

- සමාන ස්කන්ධයන්ගෙන් යුතු උපපරමාණුක අංශු යුගලය කුමක් ද?
  - (1) ඉලෙක්ටෝනය හා පුෝටෝනය

(2) පුස්ටෝනය හා නියුටෝනය

(3) නියුටුර්තය හා ඉලෙක්ටුර්තය

- (3) ඉහළ කිසිවක් නොවේ.
- පහත A, B, හා C මඟින් දැක්වෙන්නේ ශාක වල භූගත කදන් වර්ග තුණකි.







ඉහත A, B, හා C පිළිවෙලින්,

- (1) ස්කන්ධ ආකන්දය, බල්බය, කෝමය්
- (3) බල්බය, රයිසෝමය, ස්කන්ධ ආකන්දය.

- (2) කෝමය, බල්බය, රයිසොමය
- (3) රයිසොමය, ස්කන්ධ ආකන්ධය, කෝමය
- 5. යකඩ නිස්සාරණයේදී භාවිතා කරන අමුදුවරයක් නොවන්නේ මින් කුමක් ද?
  - (1) omodes!

(2) BEORS

- (3) someoc!
- (4) and Bano
- 6. ජීවීන්ගේ වියළි බරෙහි සිදුවන අපුත්වර්ත වැඩිවීම කුමක නමකින් හැඳින් වේ ද?
  - (I) වර්ධනය

(2) Description

- (3) සමාපයේජනය
- (4) අනුවර්ථනය
- පහත දී ඇති කර<del>ංග වර්ග අතුරින්</del> විදයුත් ඩුම්බක කරංග ආකාරයක් සඳහා උදාහරණයක් වන්නෙ.
  - (1) භූතාමපත කරංග
- (2) m部D からつの

- (3) දෘගය ආලෝකය (4) ජල කරංග
- 8. ජල අණුවක සාපේක්ෂ අණුක ස්කන්ධය 18 කි. මේ අනුව පහත දැක්වෙන පුකාශ අතුරින් නිවැරදි පුකාශය ment c?
  - ජල අණු මඩුලයක ස්කන්ධය 18 g වේ.
  - (2) ජලය මඩුලයක ජල පරමාණු 18 ක් අන්තර්ගත වේ.
  - (3) ජලය I g ක් තුල ජල අණු 6.022 × 10<sup>23</sup> ක් අත්තර්ගත වේ.
  - (4) ජලය මඩුල 1 ක් කුළ ඔක්සිජන් හා හයිඩුජන් මඩුල 18 බැගින් අන්තර්ගත වන බවයි.

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c experiment 3 lab report chemistry lab 117 experiment 3 pressure - Aug 08 2023

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16.0 c 289 k 0 33 97 90 kpa 19.0 c 292 k 0 34 98 97 kpa 99 22 kpa <math>101.4 kpa 19.1 c

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts