

**Unit 1 Progress Check: MCQ**

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1. Which of the following numerical expressions gives the number of moles in 5.0 g of CaO?

(A)  $5.0 \text{ g} \times 56 \text{ g/mol}$

(B)  $\frac{5.0 \text{ g}}{56 \text{ g/mol}}$

(C)  $\frac{56 \text{ g/mol}}{5.0 \text{ g}}$

(D)  $\frac{1}{5.0 \text{ g}} \times \frac{1}{56 \text{ g/mol}}$



2. In a lab a student is given a 21 g sample of pure Cu metal. Which of the following pieces of information is most useful for determining the number of Cu atoms in the sample? Assume that the pressure and temperature in the lab are 1.0 atm and 25 °C.

(A) The molar mass of Cu

(B) The density of Cu at 25 °C

(C) The volume of the Cu sample

(D) The ratio of the two main isotopes found in pure Cu



3. A 1.0 mol sample of which of the following compounds has the greatest mass?



# Science Lab Progress Check Unit Cquestions

**Christian G. Meyer**



**Science Lab Progress Check Unit Cquestions:**

## Decoding **Science Lab Progress Check Unit Cquestions**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Science Lab Progress Check Unit Cquestions**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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