

MANUAL HANDLING

1. STOP AND THINK

- Where is the load going?
- Are there any obstacles?
- Are there any aids to help me?
- Consider the size, shape and weight of the load.

2. WHILE LIFTING

- ☑ Keep distances to a minimum
- ☑ Prepare the area (remove trip hazards)
- ☑ Test the load
- ☑ Feet apart
- ☑ Leading leg slightly forward
- ☑ Hips and knees relaxed
- ☑ Firm hold from below
- ☑ Lead with your head as you lift
- ☑ Keep movement smooth
- ☑ Keep load close to your body
- ☑ Reverse the procedure when lowering the load.

3. DON'T

- ☒ Jerk overstretch
- ☒ Twist
- ☒ Lift loads which are too heavy

4. ASK FOR HELP

If the load is too heavy or you have a problem, ask for help.



Safety Poster In Manual

O García



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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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questions write the balanced equation

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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