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now using the half life equation in example 1 we have

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web solution if we compare the time that has passed to the isotope s half life we note that 44 0 s is exactly 4 half lives so using the previous equation n 4 substituting and solving results in the following less than one third of a gram of 20 f remains skill building exercise the half life of 44 ti is 60 0 y

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web solution 7 2 2 4 3 half lives 1 2 0 125 the amount remaining after 3 half lives 100 0 g x 0 125 12 5 g remaining problem 2

pd 100 has a half life of 3.6 days if one had  $6.02 \times 10^{23}$  atoms at the start how many atoms would be present after 20.0 days  
solution 20.0 / 3.6 = 5.56 half lives

**half life wjec test questions bbc** - Feb 11 2023

web radioactive isotopes are used for blood flow monitoring cancer treatment paper mills carbon dating and smoke alarms  
each isotope used in these applications has a characteristic half life