AQA Chemistry GCSE Student Book answers

C1.1 Atomic structure

Guestion	Answer	Marks	Guidance
1 a	metals: Ba, V, Hg, K, U; non-metals: P, Kr	1	
1.0	non-metal. If does not conduct electricity	11	
2	mixture: atoms not chemically bonded, compound: atoms chemically bonded	1	
3	element: all atoms same type, compound: more than one type of atom chemically combined	1	
•	tiny central nucleus surrounded by orbiting electrons	100	
5 8	madroum	1.1	
5 b	Burum	1.7	
5.0	plumbum	1.10	
5 d	Rafficers	11.	
6	twice as many oxygen atoms as carbon atoms? carbon and oxygen atoms in carbon dioxide bond to each other in the ratio 1:2 (carbon: oxygen)	2	

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This resource sheet may have been sharped from the arginal.

C1 Atomic structure

AQA Chemistry GCSE Student Book answers

C1.2 Chemical equations

Question	Answer	Marks	Guidance	
7.6	No atoms are created or destroyed in a chemical reaction so equal numbers of each type of atom must appear on both sides of the equation.	1		
1 b	H ₀ + CI ₂ → 2HCI	1		
2.8	17.6 g	1		
2 a 2 b	MgCO ₂ (s) → MgO(s) + CO ₂ (g)	1.0		
3 8	2KNO ₃ → 2KNO ₃ + O ₃	1.5		
3 b	4Li + O ₁ → 2Li ₂ O	1.1		
3.6	4Fe + 3O ₂ → 2Fe ₂ O ₃	1.0		
3 d	Fe ₂ O ₃ + 3CO 2Fe + 3CO ₃	11		
4	2Na(ii) + 2H ₂ O(i) 2NaOH(aq) + H ₂ (q)	22		

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Michael Seilmaier

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