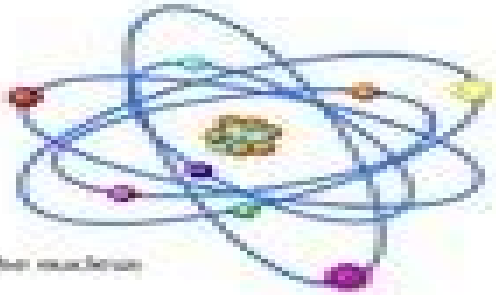
6.4 Matter Study Guide



Vocabulary:

atom- the basic building block of all matter
matter- anything that has mass and takes up space
period- the horizontal rows on the periodic table
group- the vertical rows on the periodic table
protons- positively charged particles inside the nucleus
electrons- particle in the most energetic and moves rapidly around the outside of the nucleus
nucleus- center of an atom; contains protons and neutrons
atomic number- how many protons are in an atom
molecule- atoms that are formed by two nonmetals
element- a substance cannot be broken down chemically into other substances
compound- consists of two or more substances that are all chemically combined
mixture- a substance made of two or more materials that can be separated back into its original materials
solution- a mixture that forms when one substance dissolves another
chemical bond- the force that holds atoms together
chemical change- a change in matter that produces a new substance
physical change- breaking or mixing substances
chemical equation- a shorter, easier way to show chemical reactions, using symbols instead of words

Groups

Periods

Periodic Table of the Elements

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Clifford Lane

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