

20. You have two solutions, A and B. The pH of solution A is 6 and pH of solution B is 8. Which solution has more H^+ ion concentration ?

- (1) A
- (2) B
- (3) Both have equal concentration
- (4) Cannot be determined.

21. An aqueous solution turns red litmus solution blue. Excess addition of which of the following solutions would reverse the change ?

- (1) Baking powder
- (2) Lime
- (3) Ammonium hydroxide solution
- (4) Hydrochloric acid

22. When a mixture of iron oxide and aluminium powder, is heated, intense heat is generated. This fact is made use of in

- (1) Soldering
- (2) Welding of steel
- (3) Making iron
- (4) Cutting of metals

23. Which of the following statements is/are correct about iron?

- (A) Stainless steel is an alloy of nickel, chromium and iron.

25. How are the following related to each other ?

- (A) 3-Methylbutan-2-ol
- (B) 3,3-Dimethylbutan-2-ol
- (C) 2,3,3-Trimethylbutan-2-ol
- (1) They are allotropes.
- (2) They are isomers.
- (3) They are homologues.
- (4) They are isobars.

26. An element X has electronic configuration 2, 8, 3. What is true about this element ?

- (A) Its atomic radius will be more than that of an element Y having electronic configuration 2, 8, 7.
- (B) The oxide formed by X will react with aqueous solution of hydroxide of an element with electronic configuration 2, 8, 1.
- (C) It lies in the same group as Boron (metalloid) and gallium.

- (1) A & B
- (2) B & C
- (3) Only B
- (4) A, B & C

27. Which of the following properties generally increases on moving from top to bottom in a group ?

- (1) Ionization energy
- (2) Non-metallic character

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