

SAMPLE PAPER - 3

Time : 3½ Hrs.

MM : 100

GENERAL INSTRUCTIONS

- I. Question paper comprises of three sections, A, B and C you are to attempt all the three sections.
- II. All questions are compulsory.
- III. All questions of section A, section B and section C are to be attempted separately.
- IV. Question numbers 1 to 3 in section A are one mark each, to be answered in one word or one sentence.
- V. Question numbers 4 to 7 are two marks each, to be answered in about 30 words.
- VI. Question numbers 8 to 19 are three marks each, to be answered in about 50 words.
- VII. Question numbers 20 to 24 are five marks each, to be answered in about 70 words.
- VIII. Question numbers 25 to 42 in section B are MCQs based on practical skills. Each question is a one mark question.
- IX. Question numbers 43 and 44 in section C are OTBA based questions. Each question is of five marks. The material for these questions has been provided on pages (A-1-A-8 & A-19-A-24) respectively.

SECTION - A

1. Select the isotopes form the following?

$^{21}_{10}\text{A}$; $^{22}_{10}\text{B}$; $^{23}_{11}\text{C}$; $^{23}_{12}\text{D}$

2. Which simple tissue is used for making ropes?
3. What is the amount of work done on a body moving in a circular path?
4. What are the conventions followed for writing the scientific names?
5. Name the viral disease which is about to be completely eradicated from the world. What are its preventive measures?
6. Why is it difficult to hold a school bag having a strap made of a thin and strong string?
7. A force of 7 N acts on an object. The displacement is say 8m in the direction of the force. Let us take it that the force acts on the object through the displacement. What is the work done in this case?



8. (a) State Bohr-Bury rules for the distribution of electrons in different energy levels.
(b) Calculate the maximum number of electron that can be accommodated in p shell.

9. (a) A compound of nitrogen and oxygen has the formula NO . In this compound there are 1.143g of oxygen for each 1.000g of nitrogen. A different compound of nitrogen and oxygen has the formula NO_2 . How many grams of oxygen would be combined with each 1.000g of nitrogen in NO_2 ?

(b) How many moles of nitrogen atoms are combined with 8.60 mol of oxygen atoms in dinitrogen pentoxide, N_2O_5 ?

10. Daman a student of modern school has to do a home assignment on mole concept. He was confused about total number of oxygen atoms present in one mole of O_2 and O_3 . He asked his friends Mohan and Rahul for help. Mohan said that O-atoms in 1 mole of O_3 is greater whereas Rahul argued that number of O-atoms in both will be equal.

- (a) Whose answer is correct and why?
- (b) Write the values displayed by the one with correct answer.
- (c) How many moles of sulphur atoms and oxygen atoms are present in one mole each of H_2SO_4 , H_2SO_3 and SO_2 ?

11. Which group-Pisces or Amphibia is advanced? Give two reasons.

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N Noddings



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