

DAY — 07

SEAT NUMBER /

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2023 II 28

1100

J-241

(E)

SECRETARIAL PRACTICE (52)

Time : 3 Hrs.

(4 Pages)

Max. Marks : 80

- Note :*
- (1) All questions are compulsory.
 - (2) Figures to the right indicate full marks for the questions.
 - (3) Figures to the left indicate question numbers.
 - (4) Answer to every question must be started on a new page.

- Q. 1.** (A) Select the correct answer from the options given below and rewrite the sentences : (5) [20]
- (1) Company has to pay _____ to government.
(a) taxes (b) dividend (c) interest
 - (2) _____ shares are issued free of cost to existing equity shareholders.
(a) Equity (b) Right (c) Bonus
 - (3) _____ is a proof of title of shares.
(a) Register of member
(b) Share certificate
(c) Letter of Allotment
 - (4) Debenture Capital is a _____ capital of a company.
(a) owned (b) permanent (c) borrowed
 - (5) Dividend is paid first to _____ shareholders.
(a) equity (b) preference (c) deferred

0 2 4 1

Nequivalent Qeustion Paper 24

Clifford Lane



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web use the sim to learn about isotopes and how abundance relates to the average atomic mass of an element sample
learning goals define isotope using mass number atomic number number of protons neutrons and electrons given information about an element find the mass and name of an isotope

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web use the link phet colorado en simulation isotopes and atomic mass to go to the phet website for the simulation isotopes and atomic mass begin with the isotopes and atomic mass click on the symbols next to the bar with symbol and abundance in

nature to display the data

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web multiple choice 1 minute 1 pt lithium has only two stable isotopes use the sim to determine the average atomic mass of a sample containing three lithium 6 atoms and two lithium 7 atoms 6 51556 amu 6 61565 amu 6 34875 amu 6 41548 amu

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web write at least three conclusions from your exploration 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3 the more unstable the atom the less presence or no presence in nature 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3

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web isotopes and atomic mass model 1 make isotopes open the isotopes and atomic mass simulation play with the make isotopes tab of the simulation for a few minutes and then answer the following questions 1 what particles determine the mass number number of protons and neutrons 2 why is mass number always a whole

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web this guided inquiry activity has two learning goals students will be able to students will be able to 1 explain the difference between atomic mass and mass number and 2 calculate average atomic mass from percent abundance and isotopic mass

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web find the average atomic mass of an element given the abundance and mass of its isotopes predict how the mass and name of an isotope will change given a change in the number of protons neutrons or electrons

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