

SNC2D Chemistry Unit Test Practice

Multiple Choice (1 mark each):

- Which of the following will form positive ions?
(A) the alkali metals (B) the halogens (C) the noble gases (D) all of the above
- Which of the following compounds is an ionic compound?
(A) CO_2 (B) CaO (C) CH_4 (D) all of the above
- Which of the following compounds is a molecular compound?
(A) PCl_3 (B) PbCl_2 (B) $\text{Pb}(\text{ClO}_3)_2$ (D) all of the above
- The Roman numeral in the name iron (III) nitride indicates:
(A) the number of iron atoms in the compound
(B) the number of nitrogen atoms in the compound
(C) the valence of the iron
(D) the valence of the nitrogen
- The bond in a molecule of Cl_2 could be described as:
(A) covalent (B) ionic (C) both (A) and (B) (D) neither (A) nor (B)
- If a strip of magnesium reacts with oxygen gas, the mass of the resulting magnesium oxide will be _____ the mass of the magnesium.
(A) less than (B) more than
(C) the same as (D) It cannot be determined.
- In an equation representing a chemical reaction, the reactants are always written:
(A) to the left of the arrow (B) to the right of the arrow
(C) above the arrow (D) below the arrow
- Which of the following is an indication a chemical reaction has occurred?
(A) the formation of a precipitate (B) a colour change
(C) a temperature change (D) all of the above
- If AlBr_3 reacts with Cl_2 , the products of the reaction will be:
(A) AlCl_2 and Br_2 (B) AlCl_3 and Br_2 (C) Al and Br_2Cl_2 (D) Al and BrCl
- The reaction in Question 9 could be described as a _____ reaction.
(A) decomposition (B) double displacement
(C) single displacement (D) synthesis

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