Chapter 19 - Nuclear Chemistry

This packet contains background and explanatory information on Nuclear Chemistry in addition to practice handouts. The information should be read and used to complete all of the practice included. I will be collecting this packet and giving points for the completion of the contents. You may want to consult your textbook (Chapter 19) or online resources for addition explanations and examples if needed.

Chapter Objectives: The student will (be able to)...

- Compare and contrast alpha, beta, and gamma radiation.
- Identify alpha decay, beta decay, fission, and fusion reactions.
- Balance nuclear equations.
- · Solve half-life calculations.

Part I: Radioactivity - What is it? http://www.darvill.clara.net/mucrad/index.htm

All substance are made of atoms. These have electrons (e) around the outside, and a nucleus in the middle.

The nucleus consists of protons (p) and neutrons (n), and is extremely small. (Atoms are almost entirely made of empty space!)

In some types of atom, the nucleus is unstable, and will decay into a more stable atom. This radioactive decay is completely spontanceus. The energy that is released from the nucleus of the atom is radiation.

When an unstable nucleus decays, there are three ways that it can do so. It may give out:-

- an alpha particle (we use the symbol α)
- a beta particle (symbol β)
- a gamma ray (symbol γ)

Many radioactive substances emit $\alpha_{\text{particles and}} \beta_{\text{particles as well as}} \gamma_{\text{rays}}$. In fact, you won't find a pure $\gamma_{\text{source: anything that gives off}} \gamma_{\text{rays will also give off}} \alpha_{\text{and/or}} \beta_{\text{too.}}$

A. Alpha Particles - More Information

- Alpha particles are made of 2 protons and 2 neutrons.
- This means that they have a charge of +2, and a mass of 4 (the mass is measured in "atomic mass units", where each proton & neutron=1)

We can write them as ${}^{4}_{2}$ Ct, or, because they're the same as a helium nucleus.

- Alpha particles are relatively slow and heavy.
- They have a low penetrating power you can stop them with just a sheet of paper.
- Because they have a large charge, alpha particles ionize (pull electrons from) other atoms strongly.



Nuclear Chemistry Homework Packet Key

Yan Bai

Nuclear Chemistry Homework Packet Key:

General Chemistry Workbook Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and guestions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 guestion practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material **NUCLEAR CHEMISTRY** NARAYAN CHANGDER, 2024-05-16 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel https www youtube com smartquiziz I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today s academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests guizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams guizzes trivia and more Chemistry Homework Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more College Chemistry II Norman H. Nachtrieb, 1960 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **EBOOK: GENERAL** CHEMISTRY, THE ESSENTIAL CONCEPTS CHANG, 2013-01-07 EBOOK GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS Resources in Education ,1994-04 **Curriculum Materials for Official Gazette** Philippines, 1993

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