

Indicate the type of reaction (or NR for no reaction,) complete the word equation, and write a balanced chemical equation for each of the following. Make sure to include phase symbols. ** include state of matter*

SD 1. Aluminum metal and hydrochloric acid →



DD 2. Calcium hydroxide and nitric acid →



x NR 3. Aluminum and Magnesium →

S 4. Mercury(II) and oxygen → *cannot 2 more solids, both tend to form cations*



DD 5. Zinc chloride and hydrogen sulfide →



NR 6. Silver chloride and sodium nitrate →

D 7. Sodium chlorate is heated → *both products are aqueous
@ least 1 product must be a precipitate, H₂O, or a gas*



DD 8. Barium nitrate and sodium chromate →



DD 9. Sodium bromide and silver nitrate →



DD 10. Calcium phosphate and aluminum sulfate →



D 11. Zinc carbonate is heated →



S 12. Potassium and fluorine →



S, C 13. Hydrogen and oxygen →



NR 14. Lead(II) and potassium chlorate →



S 15. Sodium and phosphoric acid →



Nuclear Chemistry 20 Answers

SA Adler



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An Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

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General and Inorganic Chemistry in Medicine Budin Michov,2024-12-09 This book is intended for students in medicine pharmacy and dentistry physicians dentists pharmacists biochemists and more In General Chemistry the laws of chemistry the structure of simple and complex compounds chemical bonds solutions chemical reactions kinetics equilibrium thermodynamics protolytic and redox processes and sorption are discussed In Inorganic Chemistry chemical elements inorganic compounds and their significance for medicine are presented It is focused on developing metal based diagnostic and therapeutic agents The significance of coordination chemistry to modulate enzyme activity is discussed The production of reactive oxygen species selectively damaging cancer cells is described too Short biographies of chemists and scientists which have rendered services to general and inorganic chemistry in medicine are given in a person index

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