

Nuclear Chemistry Half-Life Calculations

Notes and Worksheet

Includes:

- Student Note Page
- 3 ways for students to complete practice problems
- Exit Ticket

Name: _____ Date: _____

Half-Life Worksheet

Directions: Review the examples and show your work on each calculation.

1. Sodium-24 has a half-life of 15 hours.

a. How many half-lives will have occurred after 45 hours?

$$\frac{\text{Time passed}}{\text{Length of half-life}} = \frac{45 \text{ hours}}{15 \text{ hours}} = 3 \text{ half-lives}$$

b. How many grams of Sodium-24 will be left after 45 hours if we started with 40 grams?



2. Iridium-203 has a half-life of 4.5 billion years.

a. How many half-lives will have occurred after 13.5 billion years?

b. How many grams of a 30g sample of Iridium-203 will be left after 13.5 billion years?

Nuclear Chemistry Half Life Pogil Answer Key

Robert Ornduff



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