



Unit: Nuclear Chemistry "Half-Life" - Worksheet



Directions: Answer the following questions to the best of your ability. Write your answers in complete sentences and show your work where appropriate.

1. What determines the type of decay a radioactive will undergo?

The number of protons and the number of neutrons in the nucleus.

2. What is half-life?

The amount of time it takes for half of a substance to decay.

3. How much of a sample of radioactive material will remain after one half-life? After two half-lives?

One-half

One-fourth

One

radioactive material is available at 8 days. How much of a 20-gram sample of this radioactive material will still be left at 16 days?

10 grams

5 grams

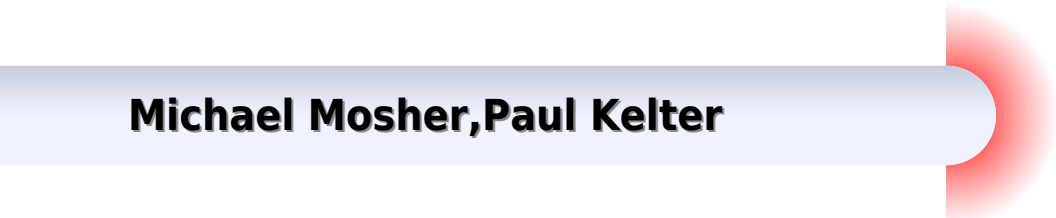
2.5 grams

4. That mass of radioactive material is found to have decreased from 100g to 6.25g in a period of 24 days. From this information, calculate the half-life of this isotope.

6 days

Nuclear Chemistry Half Life Answers

Michael Mosher, Paul Kelter



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