

25

NUCLEAR CHEMISTRY

SECTION 25.1 NUCLEAR RADIATION (pages 799–802)

This section describes the nature of radioactivity and the process of radioactive decay. It characterizes alpha, beta, and gamma radiation in terms of composition and penetrating power.

► Radioactivity (pages 799–800)

1. Which French chemist noticed that uranium salts could fog photographic plates, even without being exposed to sunlight?

2. What name did Marie Curie give to the process by which materials give off rays capable of fogging photographic plates? _____
3. An isotope that has an unstable nucleus is called a(n) _____.
4. Complete the table below to show basic differences between chemical and nuclear reactions.

Type of Reaction	Is Nucleus of Atom Changed?	Is Reaction Affected by Temperature, Pressure, or Catalysts?
Chemical		
Nuclear		

5. Complete the flowchart below, which describes the radioactive decay process.

The presence of too many or too few _____ relative to protons leads to an unstable nucleus.



At some point in time, an unstable nucleus will undergo a reaction and lose energy by emitting _____.



During the process of radioactive decay, an _____ radioisotope of one element is transformed eventually into a _____ isotope of a different element.

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1. Understanding the eBook Nuclear Chemistry Answer Key Pearson
 - The Rise of Digital Reading Nuclear Chemistry Answer Key Pearson
 - Advantages of eBooks Over Traditional Books
2. Identifying Nuclear Chemistry Answer Key Pearson
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nuclear Chemistry Answer Key Pearson
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nuclear Chemistry Answer Key Pearson
 - Personalized Recommendations
 - Nuclear Chemistry Answer Key Pearson User Reviews and Ratings
 - Nuclear Chemistry Answer Key Pearson and Bestseller Lists

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 - Nuclear Chemistry Answer Key Pearson Budget-Friendly Options
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 - Nuclear Chemistry Answer Key Pearson Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Nuclear Chemistry Answer Key Pearson
 - Highlighting and Note-Taking Nuclear Chemistry Answer Key Pearson
 - Interactive Elements Nuclear Chemistry Answer Key Pearson
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nuclear Chemistry Answer Key Pearson
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nuclear Chemistry Answer Key Pearson
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Nuclear Chemistry Answer Key Pearson
 - Setting Reading Goals Nuclear Chemistry Answer Key Pearson
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Nuclear Chemistry Answer Key Pearson
 - Fact-Checking eBook Content of Nuclear Chemistry Answer Key Pearson
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

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