CHEMISTRY (2003)

1.	Nittro	gen molecule contains	1956		N=N
	A	1 σ and 1 π bond	В.	2 σ and 2π bond	
	C.	1 σ and 2π bond	D.	2 σ and 1 π bond	
2.	Real gas obey ideal gas behavior at (decrease volume)				
	Deviate from ideal gas (increase volume) option A				
	A.	Low temp and high pressure	B.	Low pressure and high temp	
	C.	Low pressure and low temp	D.	High temp and high pressure	
3.	Oxidation state of Mn in KMnO ₄ is				
	A.	+ 6	B.	+5 K**	Mn***O4-8
	C.	+ 7	D.	+ 8	
4.	Valance shell electronic configuration of an element is 3s2, 3p4. What is the valency of the element?				
	A.	+1	В.	1 (35',3P1)valence electron=3	
	C.	- 3	D.	+3 valency=number efvalence	
5.	When ethanol reacts with H ₂ SO ₄ which one of the following product is obtained?				
	Α	Alkane	В.	Alkene	
	C.	Alkyne H ₂ SO ₄		D. No	ne of these
		C2H5OH C2H4+H2			
6.	(ethene) Which one of the following elements formed coloured compounds?				
	Α	I A group	B	Transition	clement save to st-st
	J. 30	i n group	4.0		ansition or splitting)
	C.	Halogens	D.	Allkene	
7.	Molecular formula of vinegar is				
	A	C _c H _c COOH (benzoic acid)	B.	CH,COOH	
	C.	HCOOH (formic acid)	D. C:	D. C ₂ H ₃ COOH(propanoicacid)	
8.	How many no of molecules of water in 36gm of H ₂ O				
	Α.	1.2 x 10 ²⁴	В.	12×10^{24}	
	C.	1.2 x 10 ²³	D.	6.02×10^{23}	

Number of Molucules = Mass X NA/MOL.MASS 36x6.02x1023/18=12.04x10²³ =1.204x10²³

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