

CHEMISTRY UNIT 2 REVISION

Energetics:

→ How to calculate general bond enthalpy?

- Bonds broken – bonds formed = H

→ What are the products of combustion?

- Carbon dioxide and water

→ Enthalpy change of formation conditions?

- One mole
- Standard states

→ Enthalpy change of atomisation conditions?

- One mole
- Standard conditions
- One mole of gaseous molecule forms (product)

→ What is Hess's law for combustion?



→ What is Hess's law for formation?



→ How do you use Hess's law to solve equations?

- Check whether its formation or combustion and draw accordingly
- Write the enthalpies for each product or reactant according to the arrows
- Keep in mind A is the elements or compounds used, B is the desired product, C is waste product

→ How to calculate enthalpy change?

- $Q = mct$, so $H = Q / n$

→ How to increase thermal stability?

- Decrease ionic charge
- Increase ionic radius

November 2chemistry 970paper 13

David Kirk



November 2chemistry 970paper 13:

November 2chemistry 970paper 13 Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**November 2chemistry 970paper 13**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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