

6. The average weight of Ram, Lakhan and Pavan is 67 kg. If the average weight of Ram and Lakhan is 62 kg and that of Lakhan and Pavan is 68 kg, the weight of Lakhan in kg is:  
(a) 60 (b) 62 (c) 58 (d) None of the above
7. The average of 8 numbers is 18. The average of 6 of these numbers is 15. The average of the remaining two numbers is:  
(a) 30 (b) 20 (c) 27 (d) 24
8. If 30% of a length of fabric costs Rs.126, the total length of the fabric is:  
(a) 420 (b) 380 (c) 360 (d) 400
9. Mr. Chopra is four times as old as his son. Five years ago, Mr. Chopra was nine times as old as his son was at that time. The present age of Mr. Chopra is:  
(a) 36 (b) 32 (c) 40 (d) 28
10. If David's income is 33% more than that of John's, then how much percent is John's income less than that of David's?  
(a) 24.8% (b) 32.3% (c) It cannot be determined (d) None of the above
11. The original price of a Television is Rs. 9600. The price is discounted by 20% and then raised by 10%. Its new price in rupees will be:  
(a) 8624 (b) 8448 (c) 9024 (d) 8864
12. The price of tea is increased by 20%. By how much % must a lady reduce her consumption of tea so as not to increase the expenditure?  
(a) 15.5 (b) 16.67 (c) 17.25 (d) None of these
13. A student has to secure 40% of marks to pass. He gets 178 marks and fails by 22 marks. The maximum marks are:  
(a) 600 (b) 500 (c) 400 (d) 800
14. If a boy walks from his house to the school at the rate of 4 km/hr, he reaches the school 10 minutes earlier than the scheduled time. However, if he walks at the rate of 3 km/hr, he reaches 10 minutes late. The distance from the school to the house in km is,  
(a) 2 (b) 3 (c) 4 (d) 5
15. The price of an article is cut by 20%. To restore it to the former value, the new price must be increased to:

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**Lei Shi**



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web 33mol o<sub>2</sub> 6mol br<sub>2</sub> 11mol o<sub>2</sub> 18mol br<sub>2</sub> identify the limiting reactant s and excess reactant s the limiting reactant is o<sub>2</sub> since it would yield the least amount of product 18 mol br<sub>2</sub> the excess reactant is c<sub>2</sub>h<sub>3</sub>br<sub>3</sub> since its complete reaction would have yielded up to 54 mol br<sub>2</sub> think about your result

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o<sub>2</sub> o<sub>2</sub> 3 98 g mgo vs 25 2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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web if you start with 14 82 g of ca oh<sub>2</sub> c a o h<sub>2</sub> and 16 35 g of h<sub>2</sub>so<sub>4</sub> h<sub>2</sub> s o<sub>4</sub> a determine the limiting reagent b determine the number of moles of h<sub>2</sub>o h<sub>2</sub> o produced c determine the number of grams of caso<sub>4</sub> c a s o<sub>4</sub> produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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