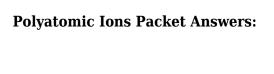
List of Common Polyatomic Ions

Cations: Symbol	Name	Charge	Symbol	Name	Charge
NH ₄ °	Ammonium	+1	Hg_{2}^{-2}	Dimercury (I)	+2
Anions:	5-4-00-0000	11.7 <u>12.4 1</u> 7.000 000 000 00			04444
Symbol	Name	Charge	Symbol	Name	Charge
$C_2H_3O_2$	Acetate	- 1	CO ₃ ·2	Carbonate	-2
HCO ₃	Bicarbonate	- 1	CrO ₄ -2	Chromate	-2
BrO ₃	Bromate	- 1	$Cr_2O_7^{-2}$	Dichromate	-2
BrO ₂	Bromite	- 1	$C_2O_4^{-2}$	Oxalate	-2
ClO ₄	Perchlorate	- 1	O_2^{-2}	Peroxide	-2
ClO ₃	Chlorate	- 1	SO ₄ -2	Sulfate	-2
CIO ₂	Chlorite	- 1	SO ₃ -2	Sulfite	-2
CIO	Hypochlorite	- 1	S ₂ O ₃ -2	Thiosulfate	-2
CN	Cyanide	- 1	SiO ₃ 2	Silicate	-2
SCN	Thiocyanate	- 1	MoO_4^{-2}	Molybdate	-2
OH	Hydroxide	- 1	WO_4^{-2}	Tungstate	-2
NO ₃	Nitrate	- 1	BO ₃ -3	Borate	-3
NO2	Nitrite	-1	PO_4^{-3}	Phosphate	-3
MnO ₄	Permanganate	- 1	PO_3^{-3}	Phosphite	-3
HSO.	Bisulfate	- 1	AsO ₄ -3	Arsenate	-3
HSO ₃	Bisulfite	- 1	AsO ₃ -3	Arsenite	-3
IO ₃	Iodate	- 1			
IO4	Periodate	- 1			
C ₆ H ₅ COO	Benzoate	-1			

Symbol	Name	Charge
$H_{2}PO_{4}^{-1}$	Dihydrogen phosphate	-1
HPO_4^{-2}	Hydrogen phosphate	-2
Fe(CN) ₆ -3	Hexacyannoferrate (III)	-3
Fe(CN) ₆ -4	Hexacyannoferrate (II)	-4

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Ken Hughes, Paul B. Kelter



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Table of Contents Polyatomic Ions Packet Answers

- 1. Understanding the eBook Polyatomic Ions Packet Answers
 - The Rise of Digital Reading Polyatomic Ions Packet Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Polyatomic Ions Packet Answers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Polyatomic Ions Packet Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Polyatomic Ions Packet Answers
 - Personalized Recommendations
 - Polyatomic Ions Packet Answers User Reviews and Ratings
 - Polyatomic Ions Packet Answers and Bestseller Lists

- 5. Accessing Polyatomic Ions Packet Answers Free and Paid eBooks
 - Polyatomic Ions Packet Answers Public Domain eBooks
 - Polyatomic Ions Packet Answers eBook Subscription Services
 - Polyatomic Ions Packet Answers Budget-Friendly Options
- 6. Navigating Polyatomic Ions Packet Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Polyatomic Ions Packet Answers Compatibility with Devices
 - Polyatomic Ions Packet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Polyatomic Ions Packet Answers
 - Highlighting and Note-Taking Polyatomic Ions Packet Answers
 - Interactive Elements Polyatomic Ions Packet Answers
- 8. Staying Engaged with Polyatomic Ions Packet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polyatomic Ions Packet Answers
- 9. Balancing eBooks and Physical Books Polyatomic Ions Packet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polyatomic Ions Packet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Polyatomic Ions Packet Answers
 - Setting Reading Goals Polyatomic Ions Packet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Polyatomic Ions Packet Answers
 - Fact-Checking eBook Content of Polyatomic Ions Packet Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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