



Science Curriculum

Chemical Changes

Read the passage and answer the following questions:



1. What is the chemical formula for hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy.

2. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula for oxygen? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in medicine as a disinfectant and in dentistry as a whitening agent. The reaction of hydrogen peroxide to form water and oxygen is a common chemical reaction that is used in many experiments and applications.

4. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, what is the mass of the products compared to the mass of the reactants?

It is equal.



5. What is the chemical formula for water? Is it a liquid, solid, or gas?

6. What is the chemical formula for oxygen? Is it a reactant or a product?

7. What is the chemical formula for hydrogen peroxide? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in medicine as a disinfectant and in dentistry as a whitening agent.

8. What is the chemical formula for water? Is it a liquid, solid, or gas?

9. What is the chemical formula for oxygen? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in medicine as a disinfectant and in dentistry as a whitening agent.

10. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, what is the mass of the products compared to the mass of the reactants?

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Lingjun Ying



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