



Reinforcement

Chemical Changes

Read the passage and answer the questions that follow.



1. What are the products of this reaction?

Water and oxygen

2. What is the reactant?

3. What is the physical state of the reactant and the products?

The reactant is a liquid, the products are a liquid and a gas.

4. According to the law of conservation of mass, if 100g of H_2O_2 is decomposed, how much H_2O and O_2 will be produced?

100g



5. What products are produced in this reaction?

6. What mass of products are produced?

7. What are the physical states of the products?

Water and oxygen

8. What is the physical state of the reactant?

9. What is the mass of the reactant?

100g

10. How much mass of products are produced?

Read the passage and answer the questions that follow.

11. A mixture of H_2 and O_2 is ignited. What are the products?

Water

Section One Reinforcement Chemical Changes

Worksheet Answers

E Durkheim



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