

Chapter 11

Use with Section 3

REINFORCEMENT**• Absolute Ages of Rocks**

Match the terms in Column I with their definitions in Column II. Write the letter of the correct phrase in the blank at the left.

Column I

- _____ 1. absolute dating
- _____ 2. half-life
- _____ 3. radioactive decay
- _____ 4. radiometric dating
- _____ 5. uniformitarianism

Column II

- a. Time it takes for half of the atoms in an isotope to decay
- b. Breaking down of a neutron into a proton and an electron
- c. Principle that Earth processes occurring today are similar to those that occurred in the past
- d. Process that uses the properties of atoms in rocks and other objects to determine their ages
- e. Calculating the absolute age of a rock by measuring the amounts of parent and daughter materials in a rock and by knowing the half-life of the parent material

Follow the steps below to demonstrate the radioactive decay of carbon-14. Then answer the questions.

1. Cut a strip of paper 8 cm long. Think of the paper as all of the carbon-14 in an animal when it died.
2. The idea is to show how you find the age of a rock that contains an animal fossil by using the half-lives of isotopes. Cut the strip of paper in half.
3. Discard one half of the paper. This represents the decayed material. Record the cut in Item 6 below with an X.
4. Continue by cutting the second half of the paper in half. Record the cut below.
5. Continue Steps 3 and 4 until the paper is so small you cannot make another cut. Record each cut you make.
6. Number of cuts: _____
7. What is the total number of times you were able (practically) to cut the sample in half?

8. Each cut represents the half-life of carbon-14. What length of time is represented by each cut?

9. Multiply the number of cuts by the half-life of carbon-14. What is the total amount of time represented by the cuts? _____
10. Could using the half-life of carbon-14 determine when dinosaurs died? Explain.

Section 3 Absolute Ages Of Rocks Worksheet

RD Boyd



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Table of Contents Section 3 Absolute Ages Of Rocks Worksheet

1. Understanding the eBook Section 3 Absolute Ages Of Rocks Worksheet
 - The Rise of Digital Reading Section 3 Absolute Ages Of Rocks Worksheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 3 Absolute Ages Of Rocks Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 3 Absolute Ages Of Rocks Worksheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 3 Absolute Ages Of Rocks Worksheet
 - Personalized Recommendations
 - Section 3 Absolute Ages Of Rocks Worksheet User Reviews and Ratings

- Section 3 Absolute Ages Of Rocks Worksheet and Bestseller Lists
- 5. Accessing Section 3 Absolute Ages Of Rocks Worksheet Free and Paid eBooks
 - Section 3 Absolute Ages Of Rocks Worksheet Public Domain eBooks
 - Section 3 Absolute Ages Of Rocks Worksheet eBook Subscription Services
 - Section 3 Absolute Ages Of Rocks Worksheet Budget-Friendly Options
- 6. Navigating Section 3 Absolute Ages Of Rocks Worksheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 3 Absolute Ages Of Rocks Worksheet Compatibility with Devices
 - Section 3 Absolute Ages Of Rocks Worksheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 3 Absolute Ages Of Rocks Worksheet
 - Highlighting and Note-Taking Section 3 Absolute Ages Of Rocks Worksheet
 - Interactive Elements Section 3 Absolute Ages Of Rocks Worksheet
- 8. Staying Engaged with Section 3 Absolute Ages Of Rocks Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 3 Absolute Ages Of Rocks Worksheet
- 9. Balancing eBooks and Physical Books Section 3 Absolute Ages Of Rocks Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 3 Absolute Ages Of Rocks Worksheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section 3 Absolute Ages Of Rocks Worksheet
 - Setting Reading Goals Section 3 Absolute Ages Of Rocks Worksheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section 3 Absolute Ages Of Rocks Worksheet
 - Fact-Checking eBook Content of Section 3 Absolute Ages Of Rocks Worksheet
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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