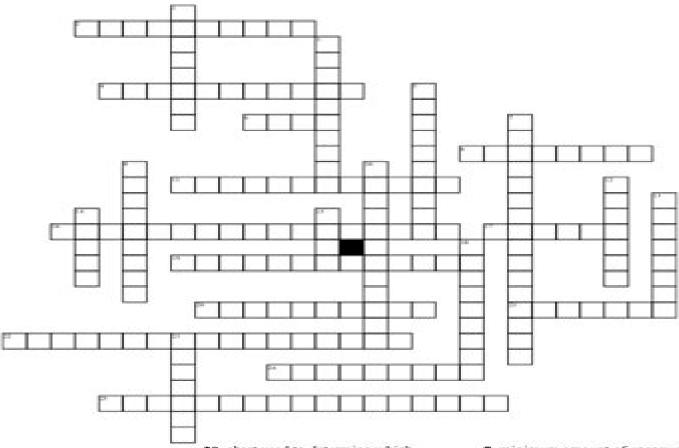
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# Chemical Equations



#### Accress

- an element will replace an element in a compound only if the single element is
- 4. solid that forms as a result of a double replacement reaction
- compound with hydroxide ion is classified as a
- substance that speeds up a chemical reaction without being permanently changed
- numbers used to balance a chemical equation
- cations switch places to form two new compounds
- 17. list of elements in order or reactivity is used with \_\_\_\_\_ replacement reactions
- in combustion reactions this gas is produced when oxygen is abundant

- chart used to determine which product is a precipitate in double replacement reactions
- proteins that acts as a biological catalyst
- 22, one element replaces another element in a compound
- 24, two or more substances combine to form a new compound
- 25. Mass is not created nor destroyed during a chemical reaction

#### Down

- chemical equation that has equal number of atoms of each element as reactants and products
- chemical reactions that release energy
   reaction of a hydrocarbon with oxygen to form water and carbon dioxide

- 7. minimum amount of energy required to start a chemical reaction
- substance that slows down or stops a chemical reaction
- 10, reaction in which one substance
- breaks down into simpler substances 12. substance in a reaction represented
- substance in a reaction represented by (aq)
- substance(s) formed as a result of a chemical reaction
- these numbers are required in compounds with transition metals
- any compound with hydrogen as its cation
- 18, substance on the left of the reaction
- 23. measure of the disorder of a system

# **Section Chemical Formulas And Equations Crossword Answers**

L Cohen

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