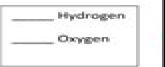
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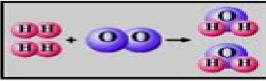
I. Chemical Changes in Matter

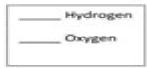
- A. Chemical Reaction
 - A change in which one or more substances are converted to ______ substances.



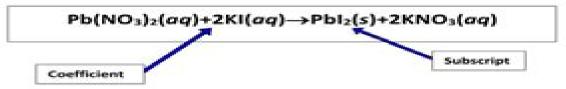
- B. Law of Conservation of Matter
 - In a chemical reaction, matter is not or ...
 - Atoms can only be ______.
 - 3. Discovered by Lavoisier.







C. Chemical Equations



- # of units of each substance.
- 2. Individual atom = atom

2Mg \Rightarrow 2 _____ of magnesium

3. Covalent substance = molecule

3CO₂ \Rightarrow 3 _____ of carbon dioxide

4. Ionic substance = unit

4MgO ⇒ 4 ______ of magnesium oxide

SYMBOL	MEANING
	produces, forms
•	plus, and
(s)	solid
Ø.	liquid
(9)	gas
(aq)	aqueous (solid dissolved in water)
Δ.	the reactants are heated

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M Tight

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