Quiz: Oxidation of Elements

Instructions: Choose the best answer for each question. Each question essesses your understanding of the oxidation of elements and related concepts.

What is exidation in terms of electron transfer?

All Gain of electrons

- 80 Loss of electrons
- Cl-Gain of protons
- D) being of neutrons
- Which of the following reactions represents oxidation?

All 20th in On in 20th Or

All Fairs in Carrier Fair

Cl. 2Na et Claire 2N act.

50 Cu + 2Agr → Cu* + 2Ag

3. In a redoc reaction, what does the term "reducing agent" refer to?

All The substance that gets oxidized

b) The substance that gets reduced

Cl The substance that remains unchanged

D) The substance that produces energy

4. Which element is most likely to be axidized in the reaction: $2n + Cu^{p_1} \rightarrow 2n^{p_2} + Cu^{p_3}$

345 The

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Ch Zhifm

100 CM

5. What is the oxidation state of sulfur in H₂SO₂?

48. +2

20 44

C) +6

6. Which of the following statements is true regarding oxidation states?

A) Oxidation states can only be positive.

88 The oxidation state of oxygen is always -2.

C) Oxidation states are always whole numbers.

Oxidation Sate Word Answer Key

C Cleary

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