

## 11.1

## DESCRIBING CHEMICAL REACTIONS

## Section Review

## Objectives

1. Represent the chemical equation
2. Balance the chemical equation
3. Use stoichiometry to calculate quantities

## Vocabulary

- |                      |                      |
|----------------------|----------------------|
| 1. chemical equation | 6. coefficient       |
| 2. stoichiometry     | 7. balanced equation |
| 3. reactant          |                      |

## Part A Completion

Write the answers to the following questions in the spaces provided. Write the balanced chemical equation for the reaction.

1. A metal carbonate reacts with hydrochloric acid to produce \_\_\_\_\_.
2. The chemical formula for a diatomic gas is \_\_\_\_\_.
3. A metal hydroxide reacts with \_\_\_\_\_ to produce water.
4. The gas produced in the \_\_\_\_\_ reaction is \_\_\_\_\_.
5. \_\_\_\_\_ reacts with \_\_\_\_\_ to produce \_\_\_\_\_.
6. \_\_\_\_\_ reacts with \_\_\_\_\_ to form \_\_\_\_\_.
7. \_\_\_\_\_ reacts with \_\_\_\_\_ to form \_\_\_\_\_.
8. \_\_\_\_\_ reacts with \_\_\_\_\_ to form \_\_\_\_\_.
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# Section Review 11 1describing Chemical Reactions

**Liying Dong**



**Section Review 11 1describing Chemical Reactions:**

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- Chapter 1: Introduction to Section Review 11 1describing Chemical Reactions
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- Chapter 4: Section Review 11 1describing Chemical Reactions in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Section Review 11 1describing Chemical Reactions. This chapter will explore what Section Review 11 1describing Chemical Reactions is, why Section Review 11 1describing Chemical Reactions is vital, and how to effectively learn about Section Review 11 1describing Chemical Reactions.
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