Chapter 8 Sulutions, Acids, and Bases

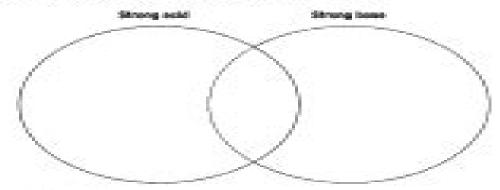
Section 8.4 Strength of Acids and Bases

(peges 246-249)

This section explains have to describe acids and bases in terms of both concentration and strength.

Reading Strategy (page 246)

Comparing and Contrasting. As you read, complete the diagram by comparing and centrasting acids and bases. For more information on this reading strategy, we the Reading and Study Skills in the Skills and Reterence Handbook at the end of your textbook.



The pH Scale (see \$47)

- II. What is the name of the number scale chemists use to describe the concentration of hydronium ions in a solution?
- 2. The pH scale ranges from _______ to ______
- 3. Circle the letter that indicates the pH of a neutral solution.
 - in. (b)
 - 150,130
 - 为了
- d. 12
- 4. Water is neutral because it contains small but equal concentrations of ______ and _____.
- Is the following someone true or false? The higher the pH value of a solution, the greater the H₂O' ion concentration is.
- If you add acid to pure water, the concentration of H₂O^{*}
 and the concentration of OH.

Section 8 4 Strength Of Acids Prentice Hall Answer

Yi-Tong Ma

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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11

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Development document for final effluent limitations guidelines and standards for the iron and steel manufacturing point source category ,2002

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