

Name \_\_\_\_\_ Period \_\_\_\_\_

# Acids and Bases Reinforcement Activity

Write "A" on the line if the following substance is an acid. Write "B" on the line if it is a base. If the substance is generally not an acid or a base, write "N" for neither. Most bases are metal hydroxides (OH), metal oxides (O<sup>2-</sup>), or ammonium (NH<sub>4</sub><sup>+</sup>) hydroxides/oxides.

1. H<sub>2</sub>O \_\_\_\_\_

2. H<sub>2</sub>SO<sub>4</sub> \_\_\_\_\_

3. NaOH \_\_\_\_\_

4. MgO \_\_\_\_\_

5. HCN \_\_\_\_\_

6. HClO<sub>4</sub> \_\_\_\_\_

7. HClO<sub>3</sub> \_\_\_\_\_

8. HClO<sub>2</sub> \_\_\_\_\_

9. SO<sub>3</sub><sup>2-</sup> \_\_\_\_\_

10. HClO \_\_\_\_\_

11. NaOH \_\_\_\_\_

12. LiOH \_\_\_\_\_

13. Ba(OH)<sub>2</sub> \_\_\_\_\_

14. NH<sub>4</sub>OH \_\_\_\_\_

15. Ca(OH)<sub>2</sub> \_\_\_\_\_

16. CaO \_\_\_\_\_

17. HNO<sub>3</sub> \_\_\_\_\_

18. C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> \_\_\_\_\_

19. HNO<sub>2</sub> \_\_\_\_\_

20. HC<sub>9</sub>O<sub>8</sub>O<sub>5</sub> \_\_\_\_\_

21. PO<sub>4</sub><sup>3-</sup> \_\_\_\_\_

22. Mg<sup>2+</sup> \_\_\_\_\_

23. H<sub>3</sub>C<sub>7</sub>O<sub>8</sub> \_\_\_\_\_

24. H<sub>3</sub>BO<sub>3</sub> \_\_\_\_\_

25. H<sub>2</sub>SO<sub>3</sub> \_\_\_\_\_

26. SrO \_\_\_\_\_

27. RbOH \_\_\_\_\_

28. Rb<sub>2</sub>O \_\_\_\_\_

29. HCl \_\_\_\_\_

30. Li<sub>2</sub>O \_\_\_\_\_

31. HBr \_\_\_\_\_

32. NaO \_\_\_\_\_

33. HI \_\_\_\_\_

34. HF \_\_\_\_\_

35. K<sub>2</sub>O \_\_\_\_\_

36. H<sub>2</sub>S \_\_\_\_\_

37. C<sub>3</sub>H<sub>8</sub> \_\_\_\_\_

38. C<sub>3</sub>H<sub>6</sub> \_\_\_\_\_

39. KMnO<sub>4</sub> \_\_\_\_\_

40. H<sub>2</sub>CO<sub>3</sub> \_\_\_\_\_

41. H<sub>3</sub>PO<sub>4</sub> \_\_\_\_\_

42. Sr(OH)<sub>2</sub> \_\_\_\_\_

Complete and balance the acid-base neutralization reactions. Write "a" under the acid (reactant), and write "b" under the base (reactant). Write "c" under the salt, which is the new ionic compound produced. Hint: carbonate ions (CO<sub>3</sub><sup>2-</sup>) will split apart to produce CO<sub>2</sub> gas and to help form water.



## Section 3 Reinforcement Acids Bases And Salts

**N Colangelo**

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