Limiting Reactant, Theoretical Yield, Actual Yield, & Percent Yield 22,99 + 35,45 = 58,44

If we have 53.2 g of Na and 65.8 g of Cl₂, what are the limiting reactant and theoretical yield?

Section 9 3 Limiting Reagent And Percent Yield Key

Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George

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