Solubility worksheet

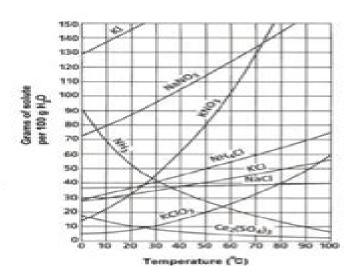
Write the correct term (or terms) in the space provided.

- 1. As temperature increases, the solubility of gases in liquids generally
- 2. When a solute dissolves and recrystallizes at the same rate, the solution is at
- 3. A substance that does not dissolve in a polar solvent is probably
- 4. To carry an electric current, a solution must contain
- 5. A solution that contains more dissolved solute than a saturated solution contains under the same conditions is called a(n)
- 6. The solubility of CO₂(g) would with increasing solvent temperature.

Based on the graph answer the following questions:

- Which compound is LEAST soluble at 10 °C?
- 2. How many grams of KCl can be dissolved in 100g of water at 80°C?
- 3. How many grams of NaCl can be dissolved in
- 100g of water at 90°C?

 4. At 40°C, how much KNO3 can be dissolved in 100g of water? _____
- 5. A saturated solution of KClO3 is formed from one hundred grams of water. If the saturated solution is cooled from 90°C to 50°C, how many grams of precipitate are formed?



What are 3 factors that affect solubility?

Section 8 2 Solubility And Concentration Worksheet Answers

OECD

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OECD Guidelines for the Testing of Chemicals, Section 4 Test No. 493: Performance-Based Test Guideline for Human Recombinant Estrogen Receptor (hrER) In Vitro Assays to Detect Chemicals with ER Binding Affinity OECD,2024-06-25 This Performance Based Test Guideline PBTG describes in vitro assays which provide the methodology for human recombinant in vitro assays to detect substances with estrogen receptor binding affinity hrER binding assays

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