# AQA Chemistry GCSE Student Book answers

#### C1.1 Atomic structure

Guestion	Answer	Marks	Guidance
1 a	metals: Ba, V, Hg, K, U; non-metals: P, Kr	1	
1.0	non-metal. If does not conduct electricity	11	
2	mixture: atoms not chemically bonded, compound: atoms chemically bonded	1	
3	element: all atoms same type, compound: more than one type of atom chemically combined	1	
•	tiny central nucleus surrounded by orbiting electrons	100	
5 a	nadrium	1.1	
5 b	Burum	1.7	
5-0	plumbum	1.10	
5 d	N. afficults	1.7	
6	twice as many oxygen atoms as carbon atoms? carbon and oxygen atoms in carbon dioxide bond to each other in the ratio 1:2 (carbon: oxygen)	2	

# Oxford University Press 2017 I were authorities and any assult inclined integerments.

This resource sheet may have been sharped from the arginal.

C1 Atomic structure

# AQA Chemistry GCSE Student Book answers

#### C1.2 Chemical equations

Question	Answer	Marks	Guidance	
7.6	No atoms are created or destroyed in a chemical reaction so equal numbers of each type of atom must appear on both sides of the equation.	1		
1 b	H <sub>0</sub> + CI <sub>2</sub> → 2HCI	1		
2.8	17.6 g	1		
2 a 2 b	MgCO <sub>2</sub> (s) → MgO(s) + CO <sub>2</sub> (g)	1.0		
3 8	2KNO <sub>3</sub> → 2KNO <sub>3</sub> + O <sub>3</sub>	1.5		
3 b	4Li + O <sub>1</sub> → 2Li <sub>2</sub> O	1.1		
3 6	4Fe + 3O <sub>2</sub> → 2Fe <sub>2</sub> O <sub>3</sub>	13.		
3 d	Fe <sub>2</sub> O <sub>3</sub> + 3CO 2Fe + 3CO <sub>3</sub>	11		
4	2Na(ii) + 2H <sub>2</sub> O(i) 2NaOH(aq) + H <sub>2</sub> (q)	22		

## **Science Lab End Of Topic Assessment C1**

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