# Atomic Structure & the Periodic Table Study Guide

#### Atomic Structure

Particle	Charge	Where are they found?
Proton	Positive	Nucleus
Neutron	Neutral/ no charge	Nucleus
Electron	negative	Electron cloud/energy levels outside nucleus

- What is the rule for writing chemical symbols like Na or Ca? Capitalize the first letter; lower case second letter
- What are the vertical columns on the periodic table called? Groups or families
  - a. How many of these are on the periodic table? 18
- b. Why are the elements in these columns? <u>They have the same chemical properties</u> (& same # of valence electrons)
  - Elements in the same group have the same number of valence electrons.
- 3. What are the horizontal rows on the periodic table called? periods
  - a. How many of these are on the periodic table?
  - Elements in these rows have the same number of energy (electron shells) levels.
- 4. What are valence electrons? Electrons in the outer energy level
- 5. How many electrons are in each energy level?
  - 1<sup>st</sup> energy level 2 2<sup>nd</sup> energy level 8 3<sup>nd</sup> energy level 8
- 6. The atomic number is equal to the # of protons and also # of electrons
- 7. The atomic mass is equal to the # protons + # of neutrons
- 8. How do you find the number of neutrons? Atomic mass atomic number
- We view the order of the periodic table today according to the atomic number

# **Periodic Table Printable Study Guide**

**Richard Bailey** 

#### **Periodic Table Printable Study Guide:**

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Guide James T. Shipman, Jerry D. Wilson, 1990

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