



Neutralization Reactions Worksheet

- 1. Write the balanced chemical equations for the neutralization reactions between:
 - a) HI and NaOH

$$HI + NaOH \rightarrow H_2O + NaI$$

b) H₂CO₃ and Sr(OH)₂

$$H_2CO_3 + Sr(OH)_2 \rightarrow 2 H_2O + SrCO_3$$

c) Ca(OH)₂ and H₃PO₄

$$3 \text{ Ca}(OH)_2 + 2 \text{ H}_3PO_4 \rightarrow 6 \text{ H}_2O + \text{Ca}_3(PO_4)_2$$

d) hydrobromic acid and barium hydroxide

e) zine hydroxide and nitrie acid

$$Zm(OH)_2 + 2 HNO_3 \rightarrow 2 H_2O + Zm(NO_3)_2$$

aluminum hydroxide and hydrochloric acid

$$AI(OH)_3 + 3 HCI \rightarrow 3 H_2O + AICI_3$$

2. Complete and balance the following equations representing neutralization reactions:

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IM Harris

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Basics of Analytical Chemistry and Chemical Equilibria Brian M. Tissue, 2023-03-02 BASICS OF ANALYTICAL CHEMISTRY AND CHEMICAL EQUILIBRIA Familiarize yourself with the fundamentals of analytical chemistry with this easy to follow textbook Analytical chemistry is the study of chemical composition concerned with analyzing materials to discover their constituent substances the amounts in which these substances are present and more Since materials exist in different states and undergo reactions analytical chemistry is also concerned with chemical equilibria the state at which various reactants and substances will undergo no observable chemical change without outside stimulus This field has an immense range of practical applications in both industry and research and is a highly desirable area of expertise for the next generation of chemists Basics of Analytical Chemistry and Chemical Equilibria provides an introduction to this foundational subject ideal for specialized courses It introduces not only the core concepts of analytical chemistry but cultivates mastery of various instrumental methods by which students and researchers can undertake their own analyses Now updated to include the latest research and expanded coverage Basics of Analytical Chemistry and Chemical Equilibria promises to situate a new generation of readers in this growing field Readers of the second edition of Basics of Analytical Chemistry and Chemical Equilibria will also find A new chapter on structure determination Revised and expanded descriptions of chemical instrumentation You try it exercises throughout to further develop practical student knowledge Compannion website of associated materials including end of chapter solutions spreadsheets for student use and more Basics of Analytical Chemistry and Chemical Equilibria is an ideal textbook for students in chemistry biochemistry and environmental science as well as students in related fields including chemical engineering and materials science for whom analytical chemistry offers a useful U.S. Government Research & Development Reports ,1969 Catalog of Instructional Films (16mm, Super 8, toolset and 8mm) for College Chemistry Advisory Council on College Chemistry, 1969 Topics-aids Rod O'Connor,1975

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