

pH Practice Worksheet

- 1) What is the pH of a solution that contains 25 grams of hydrochloric acid (HCl) dissolved in 1.5 liters of water?

- 2) What is the pH of a solution that contains 1.32 grams of nitric acid (HNO₃) dissolved in 750 mL of water?

- 3) What is the pH of a solution that contains 1.2 moles of nitric acid (HNO₃) and 1.7 moles of hydrochloric acid (HCl) dissolved in 1000 liters of water?

- 4) If a solution has a [H⁺] concentration of 4.5×10^{-7} M, is this an acidic or basic solution? Explain.

- 5) An acidic solution has a pH of 4. If I dilute 10 mL of this solution to a final volume of 1000 mL, what is the pH of the resulting solution?

Ph Practice Problems With Answers

Clemens Wendtner



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Whispering the Techniques of Language: An Mental Quest through **Ph Practice Problems With Answers**

In a digitally-driven earth wherever displays reign great and quick communication drowns out the subtleties of language, the profound strategies and psychological nuances concealed within words frequently go unheard. Yet, nestled within the pages of **Ph Practice Problems With Answers** a fascinating fictional prize sporting with fresh thoughts, lies an extraordinary journey waiting to be undertaken. Penned by a talented wordsmith, this enchanting opus invites visitors on an introspective trip, lightly unraveling the veiled truths and profound impact resonating within ab muscles fabric of each word. Within the mental depths of the touching evaluation, we will embark upon a genuine exploration of the book is primary themes, dissect their fascinating publishing type, and fail to the powerful resonance it evokes deep within the recesses of readers hearts.

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