

Stoichiometry: Practice Assessment

Name: _____

5888 - Using a balanced equation, given the grams of reactant, calc. the grams of product



Balance the equation provided. Calculate the number of moles of sodium bromide that would be produced if 25.5 moles of gallium (III) bromide are used (with an excess of sodium sulfite)

$$25.5\text{mol GaBr}_3 \times 6\text{mol NaBr}/2\text{mol GaBr}_3 = 75.75\text{mol NaBr}$$

2) Tin (II) oxide reacts with nitrogen trifluoride to produce tin (II) fluoride and dinitrogen trioxide. Using the following reaction, calculate the grams produced of each product from 50.0 grams of nitrogen trifluoride.



$$50\text{g NF}_3 \times 1\text{mol NF}_3 / 71.001\text{g NF}_3 = 0.704\text{mol NF}_3 \times 2\text{mol SnF}_2 / 1\text{mol NF}_3 \times 156.69\text{g SnF}_2 / 2\text{mol SnF}_2 = 165.5\text{g SnF}_2$$

SnF₂

$$= 0.704\text{mol NF}_3 \times 1\text{mol N}_2\text{O}_3 / 2\text{mol NF}_3 \times 76.014\text{g N}_2\text{O}_3 / 1\text{mol N}_2\text{O}_3 =$$

26.75g N₂O₃

3) A piece of copper wire is placed in silver nitrate. Silver and aqueous copper (II) nitrate is produced. The initial mass of the copper before the reaction was 2.74g and the mass remaining after the reaction was 2.32g. What is the mass of silver produced?



$$2.74\text{g} - 2.32\text{g} = 0.42\text{g}$$

$$0.42\text{g Cu} \times 1\text{mol Cu} / 63.55\text{g Cu} \times 2\text{mol Ag} / 1\text{mol Cu} \times 107.87\text{g Ag} / 1\text{mol Ag} = 1.43\text{g Ag}$$

5889 - After calculating theoretical yield in grams, calculate the percent yield. Given experimental data, determine the limiting reactant and the excess reactant

a) Through experimentation, you observe the reaction of iron (III) phosphate with sodium sulfate to make iron (III) sulfate and sodium phosphate. You began the experiment with 25g of iron (III) phosphate and 182g of sodium sulfate. How many grams of iron (III) sulfate should you have been able to produce? What is the percent yield if 18.5g of iron (III) sulfate are actually made?



$$25\text{g FePO}_4 \times 1\text{mol FePO}_4 / 150.815\text{g FePO}_4 = 0.166\text{mol FePO}_4 / 2\text{mol per rxn} = 0.083\text{rxns (LR)}$$

$$182\text{g Na}_2\text{SO}_4 \times 1\text{mol Na}_2\text{SO}_4 / 142.04\text{g Na}_2\text{SO}_4 = 1.28\text{mol Na}_2\text{SO}_4 / 3\text{mol per rxn} = 0.427\text{rxns}$$

$$0.083\text{mol FePO}_4 \times 1\text{mol Fe}_2(\text{SO}_4)_3 / 2\text{mol FePO}_4 \times 399.873\text{g Fe}_2(\text{SO}_4)_3 / 1\text{mol Fe}_2(\text{SO}_4)_3 = 33.19\text{g}$$

$$(18.5\text{g} / 33.19\text{g}) \times 100 = 56\%$$

Review Stoichiometry Section 2 Answer Key

Jin-Ying Zhang



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, DAT 2017-2018 Strategies, Practice & Review with 2 Practice Tests Kaplan Test Prep, 2016-10-04 Kaplan's DAT 2017 2018 Strategies Practice Review features the realistic practice test taking strategies and expert guidance you need to score higher on the Dental Admissions Test Our comprehensive subject review and test blueprint will help you face Test Day with confidence The Best Review Two full length online practice tests More than 600 practice questions for every subject with detailed answers and explanations 12 page tear out full color study sheets for quick review on the go A guide to the current DAT Blueprint so you know exactly what to expect on Test Day Comprehensive review of all of the content covered on the DAT Biology General Chemistry Organic Chemistry Perceptual Ability Reading Comprehension Quantitative Reasoning Kaplan's proven strategies for Test Day success Expert Guidance Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test We invented test prep Kaplan www kaptest com has been helping students for almost 80 years Our proven strategies have helped legions of students achieve their dreams OAT 2017-2018 Strategies, Practice & Review with 2 Practice Tests Kaplan Test Prep, 2016-10-04 Kaplan's OAT 2017 2018 Strategies Practice Review provides the content review test taking strategies and realistic practice you need to get the OAT results you want Updated for the latest test changes OAT 2017 2018 is your guide to facing Test Day with confidence The Best Review Two full length online practice tests More than 600 practice questions for every subject with detailed answers and explanations 16 page tear out full color study sheets for quick review on the go A guide to the current OAT Blueprint so you know exactly what to expect on Test Day Comprehensive review of all of the content covered on the OAT Biology General Chemistry Organic Chemistry Reading Comprehension Physics Quantitative Reasoning Kaplan's proven strategies for Test Day success Expert Guidance Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test We invented test prep Kaplan www kaptest com has been helping students for almost 80 years Our proven strategies have helped legions of students achieve their dreams *Holt Chemistry* Ralph Thomas Myers, 2004 **E3 Chemistry Review Book - 2018 Home Edition (Answer Key Included)** Effiong Eyo, 2017-10-20 With Answer Key to All Questions Chemistry students and homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework quizzes tests and the regents exam with E3 Chemistry Review Book 2018 With E3 Chemistry Review Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry the Physical Setting Easy to read format to help students easily remember key and must know chemistry materials Several example problems with solutions to study and follow Several practice multiple choice and short answer questions at the end of each lesson to test understanding of the materials 12 topics of Regents question sets and 3 most recent Regents exams to practice and prep for any Regents Exam This is the Home Edition of the book Also available in School Edition ISBN 978 197836229 The Home Edition contains an answer key section Teachers who want to

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Barron's Educational Series, Neil D. Jespersen, Pamela Kerrigan, 2024-07-02. Be prepared for exam day with Barron's Trusted content from AP experts. Barron's AP Chemistry Premium 2025 includes in-depth content review and practice. It's the only book you'll need to be prepared for exam day. Written by Experienced Educators. Learn from Barron's: all content is written and reviewed by AP experts. Build your understanding with comprehensive review tailored to the most recent exam. Get a leg up with tips, strategies, and study advice for exam day; it's like having a trusted tutor by your side. Be Confident on Exam Day. Sharpen your test-taking skills with 6 full-length practice tests: 3 in the book and 3 more online, plus 3 short diagnostic tests for assessing strengths and areas for improvement and detailed answer explanations for all questions. Strengthen your knowledge with in-depth review covering all units on the AP Chemistry exam. Reinforce your learning with more than 300 practice questions throughout the book that cover all frequently tested topics. Learn what to expect on test day with essential details about the exam format, scoring, calculator policy, strategies for all question types, and advice for developing a study plan. Robust Online Practice. Continue your practice with 3 full-length practice tests on Barron's Online Learning Hub. Simulate the exam experience with a timed test option. Deepen your understanding with detailed answer explanations and expert advice. Gain confidence with scoring to check your learning progress. Power up your study sessions with Barron's AP Chemistry on Kahoot! additional free practice to help you ace your exam.

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Table of Contents Review Stoichiometry Section 2 Answer Key

1. Understanding the eBook Review Stoichiometry Section 2 Answer Key
 - The Rise of Digital Reading Review Stoichiometry Section 2 Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Review Stoichiometry Section 2 Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Review Stoichiometry Section 2 Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Review Stoichiometry Section 2 Answer Key
 - Personalized Recommendations

- Review Stoichiometry Section 2 Answer Key User Reviews and Ratings
- Review Stoichiometry Section 2 Answer Key and Bestseller Lists
- 5. Accessing Review Stoichiometry Section 2 Answer Key Free and Paid eBooks
 - Review Stoichiometry Section 2 Answer Key Public Domain eBooks
 - Review Stoichiometry Section 2 Answer Key eBook Subscription Services
 - Review Stoichiometry Section 2 Answer Key Budget-Friendly Options
- 6. Navigating Review Stoichiometry Section 2 Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Review Stoichiometry Section 2 Answer Key Compatibility with Devices
 - Review Stoichiometry Section 2 Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review Stoichiometry Section 2 Answer Key
 - Highlighting and Note-Taking Review Stoichiometry Section 2 Answer Key
 - Interactive Elements Review Stoichiometry Section 2 Answer Key
- 8. Staying Engaged with Review Stoichiometry Section 2 Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review Stoichiometry Section 2 Answer Key
- 9. Balancing eBooks and Physical Books Review Stoichiometry Section 2 Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Review Stoichiometry Section 2 Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Review Stoichiometry Section 2 Answer Key
 - Setting Reading Goals Review Stoichiometry Section 2 Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review Stoichiometry Section 2 Answer Key
 - Fact-Checking eBook Content of Review Stoichiometry Section 2 Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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