

Name: _____

Unit 1 Physical Science Study Guide: Matter/Physical & Chemical Changes **Answer Key**

1. What are the three states of matter?

solid, liquid, gas

2. Name three physical properties.

color, shape, size, texture

3. If heat is removed from a bowl of water, what is eventually happen to the water?

It will freeze and become a solid.

4. When heat is added to a liquid, the liquid will melt.

5. The amount of matter in an object is its mass.

6. Jason left his bicycle outside in the rain. Over time the bicycle rusted. The rust on Jason's bicycle is an example of a chemical change.

7. A rubber duck will float on top of water because the duck has less density than the water.

8. An ice cube melting is an example of a physical change.

9. Burning a piece of paper is an example of a chemical change.

10. If you mix two substances in a chemical change, what would happen?

A new substance will form.

11. If you dissolve Kool-Aid crystals in water and then boil the water until it evaporates, the crystals will be left behind.

12. Name four ways you can tell that a chemical change has taken place.

burning takes place, bubbles forming such as fizz, rust is formed, change in color or smell

Physical Science Guided Study Work Answers Section

**United States. Congress. House.
Committee on Interstate and Foreign
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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction [The Chemical News and Journal of Physical Science](#), 1906

Reader's Guide to the History of Science Arne Hessenbruch, 2013-12-16 The Reader's Guide to the History of Science looks at the literature of science in some 550 entries on individuals Einstein institutions and disciplines Mathematics general themes Romantic Science and central concepts Paradigm and Fact The history of science is construed widely to include the history of medicine and technology as is reflected in the range of disciplines from which the international team of 200 contributors are drawn **Naval Training Bulletin**, 1944 **U.S. Naval Training Bulletin**, 1949 *Resources in Education*, 1997 **A catalogue of modern works on science and technology. 2nd, 4th, 5th, 7th, 8th, 10th-14th, 16th-19th, 22nd-25th, 35th, 39th, ed** sir George Christopher T. Bartley, 1882 **Research in Education**, 1974

Thinking About Equations Matt A. Bernstein, William A. Friedman, 2011-09-20 An accessible guide to developing intuition and skills for solving mathematical problems in the physical sciences and engineering Equations play a central role in problem solving across various fields of study Understanding what an equation means is an essential step toward forming an effective strategy to solve it and it also lays the foundation for a more successful and fulfilling work experience Thinking About Equations provides an accessible guide to developing an intuitive understanding of mathematical methods and at the same time presents a number of practical mathematical tools for successfully solving problems that arise in engineering and the physical sciences Equations form the basis for nearly all numerical solutions and the authors illustrate how a firm understanding of problem solving can lead to improved strategies for computational approaches Eight succinct chapters provide thorough topical coverage including Approximation and estimation Isolating important variables Generalization and special cases Dimensional analysis and scaling Pictorial methods and graphical solutions Symmetry to simplify equations Each chapter contains a general discussion that is integrated with worked out problems from various fields of study including physics engineering applied mathematics and physical chemistry These examples illustrate the mathematical concepts and techniques that are frequently encountered when solving problems To accelerate learning the worked example problems are grouped by the equation related concepts that they illustrate as opposed to subfields within science and mathematics as in conventional treatments In addition each problem is accompanied by a comprehensive solution explanation and commentary and numerous exercises at the end of each chapter provide an opportunity to test comprehension Requiring only a working knowledge of basic calculus and introductory physics Thinking About Equations is an excellent supplement for courses in

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