

**Chemical and Physical Properties**

State whether each of the following is a physical property or a chemical property, and type of property.

| <b>Description</b>                                                   | <b>Physical/ Chemical</b> | <b>Type</b>                            |
|----------------------------------------------------------------------|---------------------------|----------------------------------------|
| 1. The sulphur is yellow.                                            | Physical                  | Colour                                 |
| 2. Iron reacts with sulphur to give heat and a flame.                | Chemical                  | Combustibility;<br>Reaction            |
| 3. Baking soda with acid produces bubbles of gas.                    | Chemical                  | Reaction with acid                     |
| 4. Metal can be rolled into flat sheets.                             | Physical                  | Malleability                           |
| 5. Oxygen is colourless, odourless and tasteless.                    | Physical                  | Colour, odour, taste                   |
| 6. Nitrogen dioxide gas has a choking smell.                         | Physical                  | Odour                                  |
| 7. Helium is less dense than air, so a helium balloon floats.        | Physical                  | Density                                |
| 8. Wax candle burns in oxygen.                                       | Chemical                  | Combustibility                         |
| 9. An apple rots due to fungi.                                       | Chemical                  | Reaction                               |
| 10. Mercury metal is a liquid.                                       | Physical                  | State                                  |
| 11. Iodine gas is purple.                                            | Physical                  | Colour                                 |
| 12. Hydrofluoric acid is poisonous.                                  | Chemical                  | Reaction                               |
| 13. Gold is a shiny metal.                                           | Physical                  | Lustre                                 |
| 14. Cesium is the only other gold-coloured metal.                    | Physical                  | Colour                                 |
| 15. Potassium burns with a purple flame to make a white powder.      | Chemical                  | Combustibility                         |
| 16. Copper nitrate is a blue crystal.                                | Physical                  | Colour, crystal form                   |
| 17. Two colourless solutions mix to give a yellow solid precipitate. | Chemical                  | Reaction                               |
| 18. Copper can be stretched into a thin wire.                        | Physical                  | Ductility                              |
| 19. Copper metal reacts with nitric acid to make a brown gas.        | Chemical                  | Reaction with acid                     |
| 20. Salt, sodium chloride, is a white crystal that melts at 801 °C.  | Physical                  | Colour, crystal form,<br>melting point |

# Physical Properties Of Matter Answer Key

**Raffaela Di Napoli**



## **Physical Properties Of Matter Answer Key:**

## The Enigmatic Realm of **Physical Properties Of Matter Answer Key**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Physical Properties Of Matter Answer Key** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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