

2-2

Possible stereoisomers are:

a) **Eis**: **TBR**, random (because of double bond)
 b) **Isotactic**, **syndiotactic**, **atactic** (because of -CHCl- groups). However, because of distance between -CHCl- groups tacticity is unlikely to influence ability to crystallize unless rest of groups are all gls or trans.

Use $\begin{bmatrix} 1 \\ 2 \end{bmatrix}$ for $\begin{matrix} \text{C-O} = a \\ \text{O-C} = b \end{matrix}$ $\begin{matrix} 108^\circ \text{ for } \text{C-O-C} = \theta \\ 110^\circ \text{ for } \text{C-C-O} = \theta \end{matrix}$

2-3

Graphical solution is easiest using protractor and ruler. Vector summation can be used, also.



If by syndiotactic we mean a succession of **d** and **l** monomer units, the distance turns out to be the same. Another structure is possible with methyls alternating so that the repeat distance would be doubled.

2-4

Refer to Table 2-1, Fig. 2-1.
 Polychlorobutene in cyclohexane:

$$X = 0.136, \quad V_u = 108, \quad v_2 = 0.10$$

$$NV = 2.45 \times 10^3, \quad \text{so that } N = 2.27 \times 10^5 \text{ units/cm}^3$$

Same in solution:

$$\frac{X}{100} = \frac{0.557}{\text{moles in } \frac{1}{2} \text{ original volume}}, \quad V_u = 114, \quad W_u = 2.36 \times 10^{22}, \quad \text{then } v_2 = 0.21$$

2-5

$$v = 0.05, \quad \text{moles/volume} = (\text{g/vol})/(\text{g/mole})$$

If $M = M_0/M_1$, per repeat unit = 58 for $\text{C}_3\text{H}_5\text{O}$ (chain atoms)
 then $M_0 = M_1/M_1$, between cross-links = $\frac{5800 \times 58}{5} \text{ g/mole}$

$$v = 5513, \quad a = 10^{-8} \text{ meter}$$

$$V = 102 \times 1.21 \times 10^{-2} / 0.800 = 1.58 \times 10^7$$

From Fig. 2-1: $v_2 = 0.050$
 assume additive volumes. Basis: Total volume = 1 cm³

$$1.0 \text{ cm}^3 = V + V' + V_2 = 0.931 + 0.069$$

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