

1. The basic premise of the collision theory is that particles must first _____ if a reaction is to occur. Additionally, that _____ must be effective. To be effective, the particles must _____ with the _____ and the proper amount of _____.
2. _____ is the study of reaction rates which is the rate at which the _____ disappear and the products _____.
3. There are five factors affecting the reaction rate:
 - a. _____ of the _____
 - b. _____ of the _____
 - c. _____
 - d. _____
 - e. _____
4. Use the collision theory to explain:
 - a. A mixture of hydrogen gas and chlorine gas reacts faster when the volume they occupy is decreased: _____

 - b. Iron filings rust faster than iron wire: _____

 - c. An increase in temperature increases the reaction rate: _____

5. Why do you think that industrial chemists are concerned with reaction rates?

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Xiaolong Qi



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Answers to Revision Questions for Higher Chemistry ,2011 *Answers to Revision Questions for Higher Chemistry* Buchanan,2000-12-01

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