Chapter 9

9.1 Compounds and Chemical Change

- Atom- the smallest unit of an element that can exist alone or in combination with other elements.
- Compounds- formed when atoms are held together by an attractive force called a chemical bond
- Molecule- The smallest particle of a compound or a gaseous element, that can exist and still remain the characteristic chemical properties of a substance
- Chemical reaction

 a change in matter where different chemical substances are created by breaking and/or forming chemical bonds.
 - Examples: Photosynthesis
- Energy is absorbed -> produce new chemical substances -> more internal potential energy
- Energy is released when new chemical substances are produced -> less internal potential energy
- Chemical Equation: a way of describing what happens during a chemical reaction
 - Example: Photosynthesis equation-

Energy/Sunlight + carbon dioxide molecules + water molecules -> plant material molecules + oxygen molecules

9.2 Valence Electrons and Ions

- the number of electrons in the outermost orbital that usually determines the chemical properties of an atom
- <u>Valence electrons</u> the number of electrons in the outermost orbital that usually determines the properties of an atom
- Octet Rule after the element helium, atoms attempt to acquire an outer orbital with eight electrons through chemical reactions

9.3 Chemical Bonds

- chemical bond-an attractive force that holds atoms together in a compound.
- Three Classes of Chemical Bonds-
 - Ionic Bonds
 - Covalent Bonds
 - Metallic Bonds
- Fonic Bonds
 - type of linkage formed from the electrostatic attraction between oppositely charged ions in a chemical compound
- Covalent bonds: result when atoms achieve the noble gas electron structure by sharing
 electrons. They generally formed between the nonmetallic elements on the right side of
 the periodic table
- Metallic bonds: formed in solid metals such as iron, copper, and the other metallic elements that make up about 80 percent of all the elements.
- . Heat formation- energy that is released during an ionic bond formation
- Boemula- describes what elements are in the compound and in what proportions
- Simple bond covalent bond
- Double bond- formed by two pairs of electrons that are shared by two atoms

Physical Science Chapter Vocabulary

Maria C. Grant, Douglas Fisher, Diane Lapp

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<u>Developmental Cognitive Science Goes to School</u> Nancy L. Stein, Stephen Raudenbush, 2013-08-15 This book addresses core issues related to school learning and the use of developmental cognitive science models to improve school based instruction The contributors comprise a veritable who s who of leading researchers and scientists who are broadly trained in developmental psychology cognitive science economics sociology statistics and physical science and who are using basic learning theories from their respective disciplines to create better learning environments in school settings Developmental Cognitive Science Goes to School presents evidence based studies that describe models of complex learning within specific subject area disciplines focuses on domain knowledge and how this knowledge is structured in different domains across the curriculum gives critical attention to the topic of the ability to overcome errors and misconceptions addresses models that should be used to begin instruction for populations of children who normally fail at schooling This is a must read volume for all researchers students and professionals interested in evidence based educational practices and issues related to domain specific teaching and learning **Inquire, Investigate, Integrate!** Kaye Hagler, 2013-07-01 Inquire investigate integrate and inspire In this book Kaye Hagler presents thematic units that touch on core content in science with a common thread of literacy throughout The integrated units not only engage students in content such as landforms forces and motion weather life cycles and food chains but they also include reading and writing activities that engage students and connect content to literacy Options for differentiation allow for all students to access important concepts across the content areas Correlations to the NEXT Generation Science Standards and Common Core State Standards are also included for each activity

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