

Question	Answer	Marks	Guidance								
1 (a) (i)	<p>Atom(s) of an element</p> <p><b>AND</b></p> <p>with different numbers of neutrons (and with different masses) ✓</p>	1	<p><b>ALLOW</b> for 'atoms of an element':</p> <p>Atoms of the same element</p> <p>OR atoms with the same number of protons</p> <p>OR atoms with the same atomic number</p> <p><b>IGNORE</b> 'different relative atomic masses'</p> <p><b>IGNORE</b> different mass number</p> <p><b>IGNORE</b> same number of electrons</p> <p><b>DO NOT ALLOW</b> different numbers of electrons</p> <p><b>DO NOT ALLOW</b> 'atoms of elements' for 'atoms of an element'</p> <p><b>DO NOT ALLOW</b> 'an element with different numbers of neutrons' (ie atom(s) is essential)</p>								
	<p>(ii)</p> <table border="1" data-bbox="208 611 846 686"> <tr> <td>Protons</td><td>Neutrons</td><td>Electrons</td><td></td></tr> <tr> <td>74</td><td>110</td><td>74</td><td>✓</td></tr> </table>	Protons	Neutrons	Electrons		74	110	74	✓	1	
Protons	Neutrons	Electrons									
74	110	74	✓								
	<p>(iii) <math>^{12}\text{C}</math> OR C-12 OR carbon 12 OR carbon-12 ✓</p>	1	<p><b>IGNORE</b> <math>1/12^{\text{th}}</math> <b>AND</b> amu</p>								
(b)	<p>(i)</p> <p>(Oxidised):</p> <p>H (oxidation number has increased) from H = 0 to H = +1 ✓</p> <p>(Reduced):</p> <p>W (oxidation number has decreased) from W = +6 to W = 0 ✓</p>	2	<p><b>ALLOW</b> 6+ OR 6 OR 1+ OR 1</p> <p><b>ALLOW</b> one mark for correct oxidation number changes</p> <p>H = 0 to H = +1 <b>AND</b> W = +6 to W = 0</p> <p><b>ALLOW</b> oxidation states written above the equation if not seen in the text <b>BUT IGNORE</b> oxidation states written above the equation if seen in the text</p> <p><b>ALLOW</b> for one mark: (Oxidised) H has increased by 1 <b>AND</b> (Reduced) W has decreased by 6</p> <p><b>IGNORE</b> <math>\text{WO}_3</math> is reduced</p> <p><b>IGNORE</b> references to electron loss / gain if correct</p> <p><b>DO NOT ALLOW</b> incorrect references to electron loss / gain</p> <p><b>DO NOT ALLOW</b> 'H oxidised and W reduced' without reference to oxidation number changes</p>								

# Ocr F321 Jan 2013 Question Paper

**Puja Mehta**



**Ocr F321 Jan 2013 Question Paper:**

OCR(A) AS Chemistry Student Unit Guide: Unit F321 Atoms, Bonds and Groups Mike Smith, 2012-01-13 AS Chemistry

Student Unit Guide is the essential study companion for Unit F321 Atoms Bonds and Groups This book includes all you need to know to prepare for your unit exam clear guidance on the content of the unit with topic summaries knowledge check questions and a quick reference index examiner s advice throughout so you will know what to expect in the exam and will be able to demonstrate the skills required and exam style questions with graded student responses so you can see clearly what is required to get a better grade

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Ocr F321 Jan 2013 Question Paper**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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