#### Operating Manual

#### pHTestr\* 30 Pocket Testers

Thank you for purchasing an Oakton® pHTestr® Pocket pH Tester. Phease carefully read this operating manual before using the product to obtain an accurate and reliable test result and avoid unnecessary damage to the tester or sensoriprobe.

#### CE

#### Francisco en

- 3-color backdord
- 1, 2, or 5-curibration points. Auto buffer recognition: US or
- NUMBER · Stability and battery status:
- SCHOOL STREET
- Replaceable double-suriction. sensor seves your money.
- Auto power-off function: conserves battery life. IP67 waterproof nating:

- Large electro-read LCD with . No air bubbles, nonerative in blue gel inner solution.
- Accuracy of a0-01 gH a 1-digit. Buts protector can avoid. impacting. Remove it to imue. the probe.

**Bully** profestor

It's defaultsable. No rimse the produc



#### Karypaid Functions

Short press × <2 seconds. Long press × ×2 seconds.



- 1. Short press to turn on the tester and long press to turn off the tender.
- When turned off, long press to enter parameter setting.
- In measurement mode, short press to turn on bacodight.



- 1. In measurement mode, short press to switch parameter from pH to GRP (Cividation Reduction) Potential: ORP probe sold separately.
- 2. In parameter setting mode, short press to change. (harameter (unidirectional)



- Long press to entior calibration mode.
- 2. In calibration mode, short press to confern calibration. Withers measured value is looked (HOLD loom), short.
- private to surface.

#### Preparation Before Use

If it is first time use or the tester haunt been used for a longtime your some 3M KCL, solution to the Fill line in the probe cap and soulk the probe for about 15 to 30 minutes. Users can store the probe in the 3M KCL solution in the probe cap when the tester is not inuse to maintain the sensor's accuracy. To achieve maximum accuracy, we recommend society the probe in the SAFKCL storage solution-evernight (12) hours to activate the glassimoretrane thoroughty. Storing the pHO0 tester dry will NOT cause any permanent damage. It will only temporarily cause the probe to lose its sensitivity, which can always be restored by soaking in the storage solution.

#### pH Californian.

- 1. Short press-o/MEAS key to turn on.
- 2. Rinse the probe in distried water and use feaue paper to gentlydeb off excess water
- Pour pertain amount (about half volume of the calibration bottles): of pH 7 00 pH and pH 4 00 buffer solutions in the corresponding. calibration bottom.
- 4: Long press CAU+1 key to entire opionation mode, short press & RMEAS key to return to measurement mode.
- 5. Op the probe in grt 7.00 buffer solution, strippertic and allow it to stand still in the buffer solution until a stable leading is reached. When measurement stability loon (01) appears. and remains on the ecroen (see Fig 1): then short press CALINTING to complete. the one-point calibration. The tester returns to measurement mode and indication from "M" will appear at the bottom left of the soreen, indicating the risiddle point of calibration has been Astrophytical.
- 6. For 2-point palibration, start with pH 7.00 buffer edution; once confirmed then rines the probe in distilled. water and dry if with besue paper. Dio the probe into pH 4.00. buffer solution, follow the steps above to complete 2-point calibration. Indication loons: "L" and "M" will appeal on the bottom helts of the submets.
- 7. If necessary to perform a 3-point calibration, rinse probe with water, dip the probe into gHI 10.01 buffer solution (sold) separately), and follow the steps above. Indication icons 'L', 'M', and "H" will appear on the bottom left of the screen.

#### Notes

A. Tester can perform 1 to 3 points automatic caribration. Please: note that self 7.00 (LISA Standard) or self 6.86 butter existion. (NOST Standard) must be used to conduct fet point calibration. Then use other buffer solution to conduct 2nd or 3nd point. calibration. Tester will automatically recognize five kinds of pH buffer solutions. For details, please refer to the following table:

Cultivation	USA series	MIST series	Imdisoption looms	Recommended
1-point	7.00 gard	6 66 pH	88	Accuracy (d. f. ph)
2-point	7.00 pm, 4.00 pm or 1.66 pm	6.85 pH 4.91 pH or 1.68 pH	L.M	Range =7.00 pH
	7 00 pH, 10 01 pH or 12 45 pH	6.85 pH, 9.15 pH or 12.45 pH	мн	Range +7.00 pH
3-point	7 00 pH, 4 00 or 1 68 pH, 10 01 or 12 45 pH	6.95 pH, 4.01 or 1.68 pH, 9.16 or 12.45 pH	LMH	Mange. 0 to 14 00 pH

B. The buffer solutions poured into the calibration bottles are NOT for one-time use. They can be used multiple times as long as they are not contaminated, and the bottles are covered when not in use. After that, we recommend replacing the buffer solutions in the paritiration bottles with hew ones that are in the buffer bottles. (50 mL) to keep the accuracy of the standard buffer solutions. Do not your used buffer solutions back into the buffer boffes in case of contamination.

C. The tester has self-diagnostic functions.

dymitori	Self-diagnostic information	Checking and how to fix
Ee 1	Wrong pin buffer solution or the range of calibrationsolution exceeds standard.	Check whether pH buffer solution is cornect (1st point calibration must be 7 00).     Check shether the probe is demaged.     Check if there is any air bubble in the glass bulb sereor.
Ex 2	Press CALI-Taey when reading is not state during measurement.	Wast for the recoursment stability icon (0) to appear and stay, then press CALY-Yey

"The full point calibration must be 7.00 plrt. Perform the 2nd point. calibration (4.00 pH) immediately after the 1st point. Do MOT turn off the meter before you conduct 2nd point calibration. If the meter is turned off after 1st point calibration, user will need to restart the calibration process with the 7.00 gH first and the 4.00 gH following. after. Calibrating directly in pH 4.00 after turning mater off and Brack on will cause "Firth".

#### p.M. Marangar services.

- 1. Short press c-MEAS key to turn on the leater. Rinse the probe in distrived water, and pently date off excess water with tossue paper. (do not rub or wipe the glass sensor).
- 2. Dip the probe in sample solution, stir gently, and allow it to stand until a statile reading is reached. Take readings after statility. indicator icon (G) appears and slays on the soreen.

#### Section.

Name 1

A. Applications.

Model/Probe	Application		
pi-0006ulti probe	Regular water solutions' pH measurement such as hydroponics, squaculture, pools and spiss, water treatment, brewing, etc.		

#### Special Notes

- . The pH probe must be made thoroughly after each use. Soapwater should be used to clean off any grease or other contaminants.
- These pH testers will NOT give accurate or stable pH readings. when testing distilled or delonged water. This is because distilled. or deconized water do not have enough lone present for the electrode to function property. To measure distilled or delonized. water's girl, users need to use a specialized instrument technique. When testing purified water like spring water or drinking water, it. will take longer for the readings to get stabilized dypically 3 to 5. minutes) because there are very few lons left to be detected by the sensor in Trose purified water.
- Do NOT store probe in distilled or delonized purified water. because that will cause permanent damage to the pH probe. Pusitied water is only recommended for kineing the probe. The probe should be stored in 3M KCL pH electrode storage solution. ISKU 00650-040 for bent accuracy.

#### Setting the Parameters

When tester is turned off, long press 0 /MEAS key to enter parameter setting Short press MODE/A key to switch from P1 to

# **Oakton Phtestr 30 Manual**

**Aldrich Chemical Company** 

#### **Oakton Phtestr 30 Manual:**

Stoichiometry and Research Alessio Innocenti, 2012-03-07 The aim of this book is to provide an overview of the importance of stoichiometry in the biomedical field It proposes a collection of selected research articles and reviews which provide up to date information related to stoichiometry at various levels. The first section deals with host quest chemistry focusing on selected calixarenes cyclodextrins and crown ethers derivatives In the second and third sections the book presents some issues concerning stoichiometry of metal complexes and lipids and polymers architecture The fourth section aims to clarify the role of stoichiometry in the determination of protein interactions while in the fifth section some selected experimental techniques applied to specific systems are introduced. The last section of the book is an attempt at showing some interesting connections between biomedicine and the environment introducing the concept of biological stoichiometry On this basis the present volume would definitely be an ideal source of scientific information to researchers and scientists involved in biomedicine biochemistry and other areas involving stoichiometry evaluation Crosslinking of feed nutrients, microbiome and production in ruminants, 2025-08-21 The nutrition in feed and silage plays a critical role in the metabolism and health of ruminants impacting their production performance health and susceptibility to metabolic diseases like diarrhea and ruminal acidosis in dairy cows calves and goats Silage favored in dairy farms minimizes nutrient loss and facilitates easier feeding efficient feed mixing and proper handling The quantity of silage significantly influences the metabolism of dairy cows and goats The microbiome akin to a second genome facilitates the conversion of feed into milk and meat while impacting silage preparation nutrient metabolism and overall ruminant production Understanding the intricate interplay among feed nutrition microbiome and host metabolism is crucial for optimizing ruminant production This Research Topic aims to explore the synergies between feed nutrition and ruminant metabolism for enhanced production We invite submissions of perspectives reviews hypotheses methods and original research articles shedding light on the relationships among silage feed nutrition and the microbiome in dairy cows and goats Authors are invited to submit manuscripts about one or more of but not limited to the following topics Evaluation of silage quality and microbial fermentation Linkage of nutrients in feed and production of ruminants Cross talk between nutrition metabolism and the microbiome Rumen microbiota metabolism of nutrients Catalog Handbook of Fine Chemicals Aldrich Chemical Company, 1998 **InTech** .1992

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