

$$\Delta G^\circ = -236 \text{ kJ}$$

है मान निर्दिष्ट ΔG° को मान

$$\Delta G^\circ = -236 \text{ kJ mol}^{-1}$$

$$\log K_c = \frac{-\Delta G^\circ}{2.303RT}$$

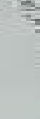
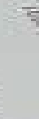
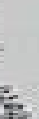
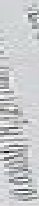
$$= \frac{(-236 \times 10^3)}{2.303 \times 8.314 \times 298} = +9.874$$

$$\text{अतः } K_c = 10^{9.874}$$

$$K_c = 32$$

5. निम्नलिखित सेले को 0.05 M का सेलर स्लोलन एवं and

निर्दिष्ट



Notes Chemistry For Up Board

Lingsheng Yao



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