

Background

- 1. Study of the _____ multi _____ relationships in a _____
- _____
- 2. Research on _____ balanced _____ equations

2. $4\text{Hg} + \text{Cl}_2 \rightarrow 2\text{Hg}_2\text{Cl}_2$

2. Magnesium reacting with a single oxygen atom produces 2 formula units of magnesium oxide

The _____ in a _____ reaction _____ the _____ _____ from the _____ involved in the _____

E.g. Problem

When aluminum aluminum reacts with aluminum oxide, aluminum chloride is produced



moles of

$\frac{1}{2}$	Al	$\frac{1}{2}$	Al_2O_3
$\frac{1}{2}$	Al	$\frac{1}{2}$	Al_2O_3
$\frac{1}{2}$	Al	$\frac{1}{2}$	Al_2O_3

2. If you start with 4 moles of Al , how many moles of Al_2O_3 will be produced?

1. $2\text{Al} + \text{Al}_2\text{O}_3 \rightarrow 2\text{AlCl}_3$ and $2\text{Al} + \text{Al}_2\text{O}_3 \rightarrow 2\text{AlCl}_3$

Note Taking Guide Episode 801

Brendan G. Carr



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