

Chemical Formula

Talking about proportions, molecular and relative of atoms indicated by a subscript.

Molecular: $C_{10}H_{12}O_4$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical: $C_5H_6O_2$ (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit.

Atomic Mass

The average mass of atoms measured relative
to hydrogen.

Isotopes

$^{12}_6C$ and $^{13}_6C$ are isotopes.

Isotopes are chemically similar
atoms.

Abundance

$^{12}_6C$ is abundant.

Isotopes are chemically similar.

$^{12}_6C$ is abundant.

Note Taking Guide Episode 6chemical Formula Key

Mike Jess



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