

### Chemical Formula

Empirical relative number of atoms of each element in compound

Molecular : covalent bonding

Formula represents the number of atoms of each

element in a single molecule

Empirical : ionic bonding

Formula represents the simplest ratio of

positive and negative ions in a formula

unit

### Electrolyte

An electrolyte that ions spontaneously ions

regardless

### Electrolyte

1. Give away electrons

2. ions are positively charged

3. Cations

### Electrolyte

1. Take electrons

2. ions are negatively charged

3. Anions

# Note Taking Guide Episode 6chemical Formula Answers

**Léo Pomar**



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